

Spatial variation in otolith elemental composition of the Pacific herring *Clupea pallasii* in northern Japan

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ABSTRACT: In order to examine whether otolith elemental composition of Pacific herring *Clupea pallasii* reflected spatially specific differences in capture location, we analyzed the elemental compositions in the edge portion of each otolith, which corresponded to the period immediately prior to the capture, as an indicator of the geographic areas in which the outer otolith was deposited. We collected 7 fish groups from 5 coastal sites: Tomamae offshore, Ishikari Bay, Akkeshi Bay, Lake Furen and Miyako Bay along the Japanese coast. Six elemental ratios, Na/Ca, Mg/Ca, K/Ca, Cu/Ca, Sr/Ca and Ba/Ca, were measured in the edge areas of each otolith by laser ablation inductively coupled plasma mass spectrometry (LA-ICP-MS). Significant differences were shown in all mean elemental ratios of otoliths among 4 sampling groups in the 2005 year class and among 3 sampling groups in the 2006 year class. The classification accuracy with the jackknife cross-validation using quadratic discriminant function analysis ranged from 80 to 98 % and 78 to 100 % in the 2005 and 2006 year classes, respectively. Our findings are comparable to those in previous studies and are further evidence that otolith chemistry is a potential tool for identifying the Pacific herring groups with different habitat-use histories and migration patterns.

KEY WORDS: Laser ablation inductively coupled plasma mass spectrometry (LA-ICP-MS) · Otolith chemistry · Pacific herring · Population structure · Movement

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INTRODUCTION

The Pacific herring *Clupea pallasii* is widely distributed in the northern part of the Pacific Ocean (Hay 1985). This species is a valuable commercial resource and an important forage species. It is therefore necessary to carry out effective fishery management based

on reliable ecological information, such as the population structure and the movement of individuals. Pacific herring is distributed in the northern parts of Japan and several populations exist with different morphologies, genetics and spawning areas (Kanno 1989a,b, Kobayashi et al. 1990, Sugaya et al. 2008). Additionally, each population has independently greatly fluctuated

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tuated in size (Iizuka & Morita 1991, Nagasawa 2001), which possibly affects the fluctuation of the total abundance of the species. In order to understand the population dynamics of Pacific herring, it is important to discriminate each ecologically different population and to determine the migration patterns for feeding and spawning in coastal and oceanic habitats.

In previous studies, the population structure and the movements of Pacific herring were examined using artificial tags and genetics (Grant & Utter 1984, Kobayashi et al. 1990, Hay et al. 2001, Sugaya et al. 2008). Artificial tagging methods however, can only link a recaptured fish to the geographic area where it was marked. Furthermore, they are impractical for small-size fishes because of the high rates of mortality during early life stages, and because of the requirement of a large number of tagged individuals to ensure a sufficient number of recaptures. Genetic composition can link fish to their origin populations, but not to the geographic areas where it was distributed. In Pacific herring, therefore, alternative approaches are required for studying the population structure and the movements of individual fish.

To date, the chemical compositions of otoliths of teleost fishes have been analyzed to elucidate ecological questions concerning their population structure (Campana et al. 1994, Milton et al. 1997), population connectivity (Gillanders 2002), migration (Limburg 1995, Secor et al. 2001, Arai et al. 2004), the onset of metamorphosis (Otake et al. 1994, Arai et al. 1997), natal origin (Ruttenberg & Warner 2006) and natal homing (Thorrold et al. 2001). Otoliths are composed of calcium carbonate crystals within a protein matrix and enable hearing and balance. They are formed prior to birth and grow with the daily deposition of a new crystal layer around the surface. Otolith elemental composition reflects environmental factors such as water temperature, salinity and elemental composition in the ambient water at the time of deposition (Townsend et al. 1992, Fowler et al. 1995, Hoff & Fuiman 1995, Bath et al. 2000, Elsdon & Gillanders 2002, Martin & Thorrold 2005). Furthermore, since otoliths are metabolically inert, the otolith elemental composition remains unvaried after deposition and preserves a record of the environmental conditions experienced by the fish. In a number of previous studies, differences in elemental compositions of otoliths have been observed among different rivers (Thorrold et al. 1998a,b), estuaries (Gillan-

ders & Kingsford 1996) and oceanic areas (Ashford et al. 2005). Therefore, the elemental composition of fish otoliths can be useful as a natural tag for determining the geographic areas or habitats that have been used by each individual fish.

The objective of the present study was to examine whether otolith elemental composition of the Pacific herring reflected spatially specific differences in capture location. We analyzed the elemental compositions in the edge portion of each otolith, which corresponds to the period immediately prior to capture and serves as an indicator of the geographic area in which the outer otolith was deposited.

MATERIALS AND METHODS

Study sites and fishes. We sampled fish from 5 coastal sites: Tomamae offshore, Ishikari Bay, Akkeshi Bay, Lake Furen and Miyako Bay (Fig. 1). Four sampling sites were in the coastal waters off Hokkaido Island, and Miyako Bay was on the northeastern coast of Honshu. All sites except Tomamae offshore were major spawning grounds for Pacific herring *Clupea pallasii*. Furen Lake included brackish sites adjoining the Okhotsk Sea. Akkeshi Bay was also affected by brackish water from the adjoining

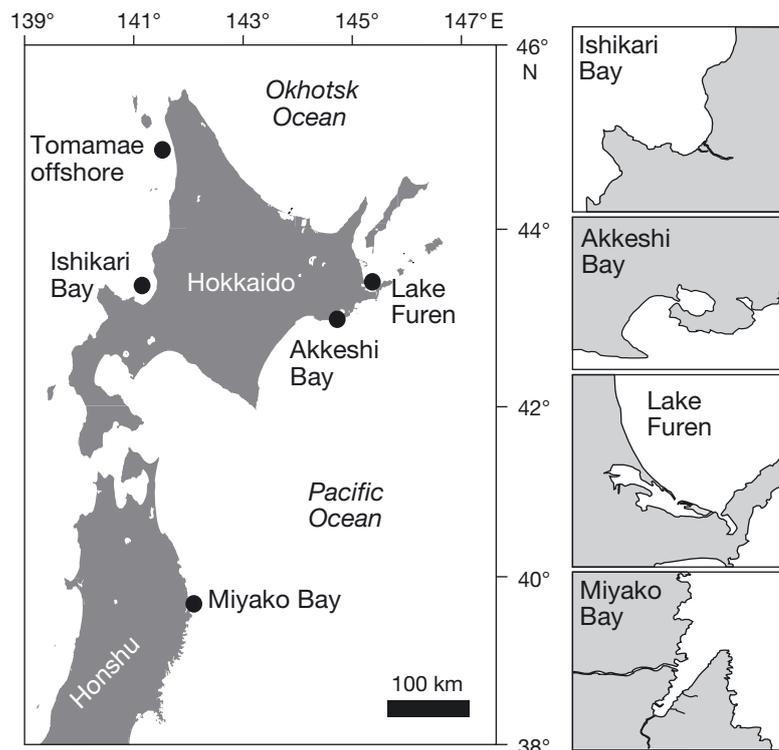


Fig. 1. *Clupea pallasii*. Location of the 5 sampling sites (●) used in the present study

Akkeshi Lake. Ishikari Bay and Miyako Bay contained mostly full-strength seawater. However, since several rivers flowed into these bays, these sampling sites were affected to some degree by riverine freshwater. There is therefore, a potential for large differences in water chemistry among the 5 sites that may cause considerable differences in otolith elemental signatures.

Adult herring were collected from the 5 sites from late January to early June 2008, which is the spawning season of Pacific herring along the Japanese coast. All sampling sites, except Tomamae offshore, are major spawning grounds for Pacific herring on the Japanese coast. Therefore, it is highly likely that the collected fish came from the geographically corresponding spawning population. In Ishikari Bay, fish from the 2005 year class were collected twice, in both March and May, and 2 year classes were examined from the Miyako Bay site. This resulted in sampling groups being obtained from 5 sites in 2008, with 1 site being sampled for the same year class in 2 different months, and 2 year classes being sampled at 1 site, resulting in 7 sampling groups from 5 locations.

For each fish, total length (TL), fork length (FL) and gonad weight were measured and subsequently the gonad-somatic index (GSI; gonad weight \times 100 body weight⁻¹) was determined. Year class was determined by counting the translucent bands in the otoliths or by counting the annuli of scales. The degree of gonad maturation was determined based on the GSI following Koya et al. (2003). Both TL and FL were measured in the 2006 year class of herring from Akkeshi Bay and Miyako Bay, but only either TL or FL was measured in the 2005 year class from Ishikari Bay in March, Ishikari Bay in May, Tomamae offshore, Miyako Bay and Lake Furen. We estimated the unmeasured values of either TL or FL of those individuals based on a linear regression analysis between TL and FL using herring from the 2006 year class in Akkeshi Bay and Miyako Bay (Fig. 2). The biological features of fish used in the present study, including sampling date, year class, TL, FL, sex and GSI, are summarized in Table 1.

Otolith preparation and chemical analysis. Both sagittal otoliths from each fish were extracted, cleaned with Milli-Q water for 5 min in an ultrasonic bath, rinsed several more times in MilliQ water, and were then air-dried and placed in a clean plastic case. Prior to chemical analysis using laser ablation induc-

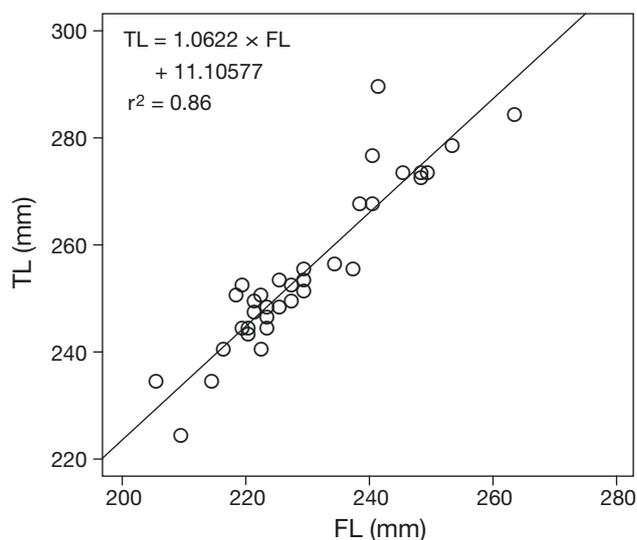


Fig. 2. *Clupea pallasii*. Results of linear regression analysis between total length (TL) and fork length (FL) using 2006 year class fish collected from Akkeshi and Miyako Bays

tively coupled plasma mass spectrometry (LA-ICP-MS), we set the otolith in the laser ablation chamber using double-sided tape without grinding and polishing.

Seven isotopes, ²³Na, ²⁴Mg, ³⁹K, ⁴³Ca, ⁶³Cu, ⁸⁸Sr and ¹³⁸Ba, were measured in the edge areas of each otolith by LA-ICP-MS. The LA-ICP-MS system used was the 7500CS ICP-MS (Agilent) coupled with the 213 nm Nd-YAG laser ablation system (New Wave Research). To improve the measurement accuracy, we applied both the He flushing technique and a stabilizer device (Tunheng & Hirata 2004). Calibration from the signal intensity to the element was performed using 4 standard materials, such as a clear calcite with known composition (Shirai et al. 2008a), NIST 612 standard glass

Table 1. *Clupea pallasii*. Specimens used for otolith elemental analysis. Sampling date, year class (YC), total length (TL), fork length (FL), number of individuals analyzed and gonad somatic index (GSI) are shown. TL, FL and GSI are presented as mean values (\pm SD). The 7 sampling groups collected from 5 sampling sites were defined as Akkeshi Bay (A), Ishikari Bay in March (I-Mar), Ishikari Bay in May (I-May), Tomamae offshore (T), Miyako Bay in January (M-Jan), Miyako Bay in March (M-Mar) and Lake Furen (F). All fishes were captured in 2008. * indicates that the values were calculated from FL (or TL) based on the relationship between TL and FL in individuals of the 2006 year class from Akkeshi and Miyako Bays (see Fig. 2)

Group	Date	YC	TL (mm)	FL (mm)	Male (n)	Female (n)	GSI (%)
I-Mar	26 Mar	2005	291.3 \pm 8.1*	263.8 \pm 7.6	6	15	20.7 \pm 4.4
I-May	29 May	2005	256.5 \pm 7.1*	231.0 \pm 7.1	12	14	16.9 \pm 3.2
T	16 Mar	2005	276.8 \pm 6.7*	250.1 \pm 6.3	4	16	15.0 \pm 1.8
M-Jan	31 Jan	2005	309.9 \pm 17.0	281.2 \pm 16.0*	2	13	21.2 \pm 3.7
A	1 Jun	2006	249.7 \pm 9.1	225.9 \pm 9.0	9	9	16.1 \pm 2.4
M-Mar	3 Mar	2006	265.5 \pm 20.5	239.0 \pm 18.8	8	11	16.3 \pm 3.4
F	24 Apr	2006	263.0 \pm 9.1*	237.1 \pm 8.5	11	10	17.4 \pm 5.3

distributed by the National Institute of Standards and Technology (USA), and pressed pellets of certified reference material of powdered coral (JCp-1) and powdered giant clam (JCt-1) distributed by the National Institute of Advanced Industrial Science and Technology (Japan) (Shirai et al. 2008b). The standard used in each analysis was chosen depending on the elemental properties of each standard material.

We made duplicate 40 μm diameter laser beam spot measurements at the edge of each otolith. The ablated areas covered about 10 d prior to the capture, since we determined 10 otolith daily rings in these areas through light microscope observations. The frequency of the laser beam was 10 Hz. The predicted crater depth was approximately 50 μm . The elemental data of each otolith were presented as the average of 2 spots. Otoliths from different sampling sites were randomly analyzed throughout the analysis. Background levels and standard references were examined before and after each scan. The isotope ^{43}Ca was used as an internal standard, and all elemental data were expressed in terms of their molar ratio to Ca. The detection limits achieved in the present study were as follows: ^{23}Na 66.24, ^{24}Mg 2.06, ^{31}P 18.73, ^{39}K 27.69, ^{63}Cu 0.16, ^{88}Sr 0.010 and ^{138}Ba 0.013 $\mu\text{mol mol}^{-1}$. These values were calculated from 3 standard deviations from the mean blank count of each isotope (gas background). All sample compositions were above the limit of detection. Mean estimates of precision (% relative standard deviation) based on JCp-1 standard were 2.67% (Na/Ca), 2.36% (Mg/Ca), 6.76% (K/Ca), 7.31% (Cu/Ca), 3.50% (Sr/Ca) and 9.35% (Ba/Ca).

Data analysis. The differences in the content of each element in the otolith among 7 sampling groups (Table 1) were tested using ANOVA and, subsequently, Tukey and Kramer's multiple comparisons among sampling groups. A multivariate analysis of variance (MANOVA) was performed to examine the geographic variations among 5 sampling groups. It was also used for examination of the monthly variation in otolith chemistry within a single sampling site using the fish groups collected from Ishikari Bay in March and May. To examine whether otolith chemistry discriminates their habitat differences, we used quadratic discriminant function analysis (QDFA) of multi-signatures for each sampling group. We tested the ability of otolith elemental compositions to discriminate habitat differences using QDFA with jackknife cross validation.

Prior to analyses of data by ANOVA and QDFA, all elemental data were examined for normality and homogeneity of variances using either Shapiro-Wilk normality tests ($\alpha = 0.05$) or Bartlett tests ($\alpha = 0.05$). To meet model assumptions, the data were transformed into natural log values of Na/Ca, K/Ca and Sr/Ca ratios or reciprocally transformed values of Mg/Ca, Cu/Ca

and Ba/Ca ratios. Normality was obtained for all 6 elemental ratios after the transformations. However, otolith Mg/Ca, K/Ca and Cu/Ca ratios were still heterogeneous in variance, and a more constructive probability of $\alpha = 0.01$ was used (Underwood 1999) for ANOVA. QDFA does not assume homogeneity of the covariance matrices; hence, QDFA was selected for classification methods.

RESULTS

We examined the difference in each elemental ratio of the edge areas of otoliths among 7 sampling groups of Pacific herring *Clupea pallasii* collected from different sampling localities and months (Table 1). Significant differences were shown in all mean elemental ratios of otoliths among 4 sampling groups in the 2005 year class, including Tomamae offshore, Ishikari Bay in March, Ishikari Bay in May and Miyako Bay in January (ANOVA, $p < 0.001$) (Fig. 3, Table 2). Tukey and Kramer's tests demonstrated that Na/Ca, K/Ca and Ba/Ca ratios of otoliths were significantly different among several sampling groups. In particular, the Ba/Ca ratios for Ishikari Bay in May were considerably higher than those for the other groups. The Mg/Ca ratios of otoliths collected from Miyako Bay in January were significantly lower than those for other sampling grounds. For the Cu/Ca ratio, higher values were found at Ishikari Bay in March and May than at Tomamae offshore and Miyako Bay in January. The Sr/Ca ratios for Ishikari Bay were significantly higher than those for the other sites. Three sampling groups of the 2006 year class, including Akkeshi Bay, Lake Furen and Miyako Bay in March, also showed significant differences for all elemental ratios of otoliths (ANOVA; $p < 0.01$) (Fig. 3, Table 3). Tukey and Kramer's tests revealed that the Na/Ca and K/Ca ratios of the otoliths were significantly different among all sampling groups. There were significant differences in the Mg/Ca ratios between Akkeshi Bay and Miyako Bay in March and between Miyako Bay in March and Lake Furen. The Sr/Ca ratios for Akkeshi Bay were significantly higher than those for Miyako Bay in March and Lake Furen. The Cu/Ca ratios were considerably higher for Lake Furen than those for the other sampling groups. The Ba/Ca ratios for Miyako Bay in March were significantly lower than those for Akkeshi Bay and Lake Furen. In multi-element fingerprints, the significant differences of otolith chemistry among sampling groups were shown in the 2005 year class (MANOVA, Pillai's trace = 1.98, $p < 0.001$) and in the 2006 year class (MANOVA, Pillai's trace = 1.45, $p < 0.001$). Furthermore, a MANOVA also showed a significant difference between fishes from Ishikari

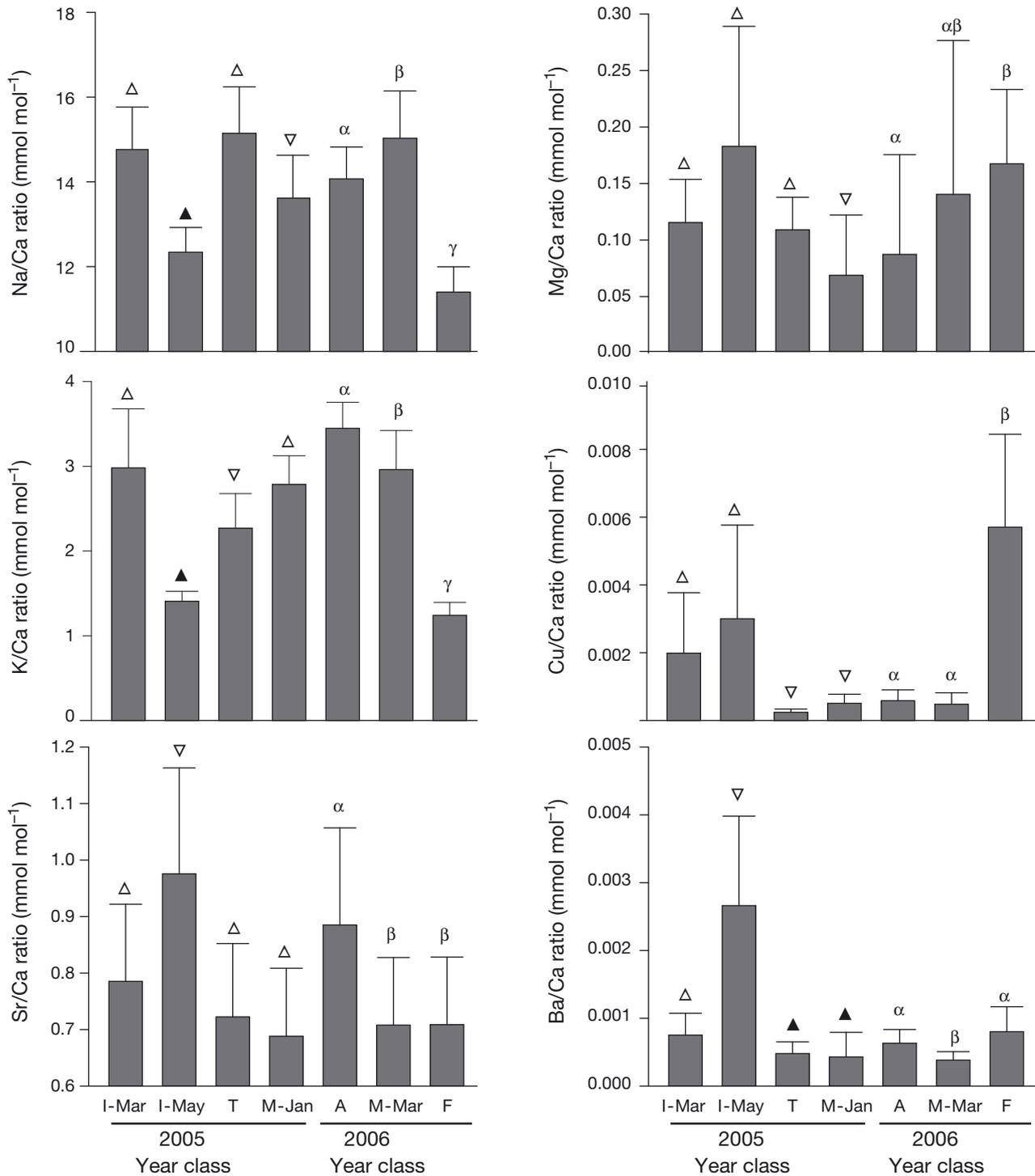


Fig. 3. *Clupea pallasii*. Otolith elemental ratios (mean + SD) of 7 sampling groups of Pacific herring: Ishikari Bay in March (I-Mar), Ishikari Bay in May (I-May), Tomamae offshore (T), Miyako Bay in January (M-Jan), Akkeshi Bay (A), Miyako Bay in March (M-Mar) and Lake Furen (F). Statistical differences are shown by different symbols (Δ, ▽ and ▲) for the 2005 year class and by Greek letters (α, β and γ) for the 2006 year class (post hoc multiple comparisons were made with Tukey and Kramer's tests)

Bay in March and May in the 2005 year class (Pillali's trace = 1.61, $p < 0.001$). Those results suggest that there are significant monthly and geographical variations in otolith chemistry in the Pacific herring.

QDFA was performed to determine the ability of elemental signature to classify adult fish to the correct localities where they were captured. In the 2005 year class, the classification accuracy examined by jack-

Table 2. *Clupea pallasii*. Results of ANOVA for each element (molar ratio to calcium) in the otoliths of 4 sampling groups (Tomamae offshore, Ishikari Bay in March, Ishikari Bay in May, Miyako Bay) of the 2005 year class

Element	Source	df	SS	F	p
Na/Ca	Site	3	0.595	48.971	<0.001
	Error	78	0.316		
Mg/Ca	Site	3	16.158	14.656	<0.001
	Error	78	28.665		
K/Ca	Site	3	7.637	95.083	<0.001
	Error	78	2.088		
Cu/Ca	Site	3	0.0338	46.759	<0.001
	Error	78	0.0188		
Sr/Ca	Site	3	1.553	15.549	<0.001
	Error	78	2.597		
Ba/Ca	Site	3	0.0222	64.646	<0.001
	Error	78	0.0089		

Table 3. *Clupea pallasii*. Results of ANOVA for each element (molar ratio to calcium) in the otoliths of 3 sampling groups (Akkeshi Bay, Lake Furen, Miyako Bay) of the 2006 year class

Element	Source	df	SS	F	p
Na/Ca	Site	2	0.846	117.94	<0.001
	Error	55	0.197		
Mg/Ca	Site	2	12.23	7.658	<0.01
	Error	55	43.916		
K/Ca	Site	2	12.148	409.79	<0.001
	Error	55	0.815		
Cu/Ca	Site	2	0.0296	118.47	<0.001
	Error	55	0.0069		
Sr/Ca	Site	2	0.566	9.172	<0.001
	Error	55	1.698		
Ba/Ca	Site	2	0.0031	16.086	<0.001
	Error	55	0.0053		

knife cross-validation ranged from 87 to 100% for the 4 sampling groups (Table 4a). When the fish from Ishikari Bay in March and May were combined, the classification accuracy changed to a range of 80 to 98% (Table 4b). Since otolith elemental compositions of fish from Ishikari Bay in March and May showed little misclassification to the other fish groups, the monthly variation would not prevent the interpretation of geographic differences of otolith chemistry among fish groups. Likewise, in the 2006 year class, the classification accuracy ranged from 78 to 100% (Table 4c).

DISCUSSION

In addition to environmental factors such as temperature, salinity and elemental compositions, factors such as ontogenetic and physiological effects have also been found to influence otolith elemental composi-

Table 4. *Clupea pallasii*. Cross-validated classification rates (%) for quadratic discriminant function analysis undertaken for 7 sampling groups taken from 5 sampling sites along the northern coast of Japan using 6 variables (Na/Ca, K/Ca, K/Ca, Cu/Ca, Sr/Ca and Ba/Ca). (a) Classification accuracies among the 4 sampling groups in the 2005 year class. (b) Results for the 3 remaining 2005 year class groups when the 2 fish groups from Ishikari Bay in March and May are combined. (c) Results for the sampling groups in the 2006 year class. Values in **bold** indicate the rates of correct classification to the captured sites and were used for the evaluation of the ability of otolith elemental compositions to discriminate habitat differences in Pacific herring. A: Akkeshi Bay; I-Mar: Ishikari Bay in March; I-May: Ishikari Bay in May; I: Ishikari Bay in March and May; M-Jan: Miyako Bay in January; M-Mar: Miyako Bay in March; F: Lake Furen; T: Tomamae offshore

(a) 2005 year class	I-Mar	M-Jan	I-May	T
I-Mar	90	5	5	0
M-Jan	13	87	0	0
I-May	0	0	100	0
T	5	5	0	90
(b) 2005 year class	I	M-Jan	T	
I	98	2	0	
M-Jan	20	80	0	
T	10	5	85	
(c) 2006 year class	A	F	M-Mar	
A	78	0	22	
F	0	100	0	
M-Mar	11	0	89	

tions in some species of fish (Kalish 1989, Otake et al. 1994, Chittaro et al. 2006). In the present study, we minimized the ontogenetic effects on the otolith chemistry by using fishes of particular year classes to examine the geographic variations among sampling groups. Physiological influences resulting from reproductive activity were also considered to be minimal, because only fish of similar reproductive development (only high GSI value) were analyzed. Further study will be needed to understand the mechanisms that may influence the elemental composition of fish otoliths. However, as reported by Campana et al. (1994) and Thorrold et al. (1998a), elemental signatures as natural tags of habitat are practical enough, even if the mechanisms of compositional control are not perfectly understood.

A monthly variation in otolith chemistry was found between Pacific herring collected in Ishikari Bay in March and May in the 2005 year class, although it did not affect the interpretation of the spatial variation in otolith chemistry. The variation appears to be due to the notable environmental changes caused by the input of melted snow into the bay via rivers during spring. Le et al. (2006) reported that the flow rate of the Ishikari

River in April was $<500 \text{ m}^3 \text{ s}^{-1}$, while the flow rate in May increased remarkably to $\sim 2000 \text{ m}^3 \text{ s}^{-1}$. Such large inputs of melted snow probably to result in drastic changes in temperature, salinity and water chemistry and may be the cause of monthly variations in otolith chemistry between months in the Ishikari Bay population. Further information on the monthly changes in water environment, and the monthly/seasonal changes in otolith chemistry of specific fish populations in relation to these environmental changes, is required.

In the present study, the classification accuracies for otolith chemistry, based on QDFA, were similar to those of other studies in which the natal or nursery sites of fish collected in estuarine and riverine systems were determined. For example, juvenile weakfish *Cynoscion regalis* were classified to their natal estuaries with 63% success (Thorrold et al. 1998a); juvenile American shad *Alosa sapidissima* were classified to their natal river with 90% success (Thorrold et al. 1998b); and juvenile trumpeter *Pelates sexineatus* were classified to their estuarine nursery habitat with 75% success (Gillanders & Kingsford 2000). Some studies have examined the ability of otolith chemistry to discriminate habitat differences in coastal or open-ocean environments. Pacific bluefin tuna *Thunnus orientalis* spawned in the Pacific Ocean were classified with 100% accuracy, but those from the East China Sea, including 3 year classes, were classified to each year class with $<50\%$ success (Rooker et al. 2001). Success in the classification of adult Patagonian toothfish *Dissostichus eleginoides* according to capture sites in a fully marine environment ranged from 50 to 84% (Ashford et al. 2005). The classification success determined in previous studies using otolith elemental compositions was comparable to that we found in the present study for Pacific herring. We conclude that otolith chemistry in Pacific herring records habitat differences among marine areas.

Our findings in the present study indicated that otolith chemistry is a potential tool for identifying groups of Pacific herring with different histories of habitat use and with different migration patterns. The information on population structure and migration patterns that can be provided by otolith chemistry is likely to be useful in helping to establish an effective fishery management strategy for Pacific herring. The results of the present study will also contribute to developing a reliable method of stock identification using otolith chemistry.

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