

Distribution and origin of dissolved DNA in lakes of different trophic states

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ABSTRACT: Concentrations and origin of dissolved DNA (dDNA) were studied in 14 lakes in the Mazurian Lake District (Northern Poland) and 7 lakes in Southern Germany during the summer stratification period. dDNA concentration varied markedly (from 0.5 to 70 $\mu\text{g l}^{-1}$) in the studied lakes. We suggest that this dissolved organic matter fraction can be potentially one of the most important P and N sources for planktonic microorganisms. Laboratory experiments and field observations suggested that eukaryotic microorganisms (algae) are the most important sources of dDNA in lake water, whereas bacteria mainly decompose the dDNA pool.

KEY WORDS: Dissolved DNA · Trophic status index of lake · Algae · Bacteria · Primary production · Secondary production

INTRODUCTION

The majority (>95%) of dissolved organic matter (DOM) in aquatic environments consists of polymeric, high-molecular-weight compounds (HMWC) (Thurman 1985, Münster & Chróst 1990). It is generally believed that the molecular weight and the size of the particles are the principal factors affecting microbial utilization of DOM (Rogers 1961, Saunders 1976, Payne 1980). Since the low-molecular-weight organic compounds (LMWC) can pass directly through the cell membranes, they are decomposed faster than HMWC and play a key role in the metabolism of aquatic microheterotrophs (Siuda et al. 1991). However, the studies carried out during the last decade made this relatively plain conceptual model more complicated. Meyer et al. (1987), Tranvik (1990) and Amon & Benner (1994, 1996) have indicated that some HMWC may be as rapidly utilized by bacteria as LMWC. These findings lead to the conclusion that knowledge of the composition and dynamics of HMWC may be equally (or even more) significant for understanding the development and activity of microbial heterotrophic communities as well as nutrients and energy cycling in natural waters.

Deoxyribonucleic acid, as a constituent common to all living cells, is a component of the dissolved macromolecular fraction (HMWC) in all aquatic environments. The relative contribution of carbon contained in the dissolved deoxyribonucleic acid (dDNA) fraction to the total DOC pool is commonly low and does not exceed 0.015%. Therefore the role of dDNA in the carbon cycle in a lake can probably be neglected. However, a comparison of the approximate atomic ratio of the main DNA components (C:N:P \approx 10:4:1; Maruyama et al. 1993) and the Redfield ratio in plankton cells (106:16:1; Harris 1986) suggests that dDNA may be an attractive source of N and P for lake microorganisms.

Ambient dDNA concentrations in aquatic environments usually vary from 10 to 40 $\mu\text{g l}^{-1}$ (Minear 1972, DeFlaun et al. 1987, Karl & Bailiff 1989, Paul et al. 1989, Maruyama et al. 1993, Weinbauer et al. 1995, Siuda & Güde 1996a) and result from a variety of biological and physicochemical processes. These processes include excretion, exudation, sloppy feeding, death and autolysis of planktonic microorganisms, enzymatic hydrolysis and sorption/desorption of dDNA by seston particles (Bailiff & Karl 1991, Ogram et al. 1994).

Three main groups of microplankton enzymes mediate the processes of liberation of the orthophosphate

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ion from a DNA molecule: DNases (endo- and exonucleases), alkaline phosphatase (AP) and 5'-nucleotidase (5'-PN). Literature information on turnover times calculated for dDNA and its enzymatic degradation products (i.e. nucleotides) (Paul et al. 1989, Ammerman & Azam 1991a, Siuda & Güde 1996b) suggests that the efficiency of DNases is the limiting step of dDNA-P regeneration in waters. The presence of DNA hydrolyzing bacteria in aquatic environments has been known for some time (Maeda & Taga 1973, 1981). The contribution of other microbial sources besides bacterioplankton to overall DNase activity is still poorly studied.

Although alkaline phosphatase is often intensively produced by both algae and bacteria during periods of orthophosphate deficiency, especially in the euphotic zone of lakes, this enzyme is probably of only minor importance in 5'-PN decomposition processes. Siuda & Güde (1994b) found that the bacterial orthophosphate resistant enzyme 5'-PN had a much higher affinity to nucleotides than AP. Moreover, Ammerman & Azam (1991b) and Thingstad et al. (1993) pointed out that the nucleotide pool is important in the turnover of organic phosphoesters. Considering that the amount of dDNA in aquatic ecosystems is relatively high in comparison to the other organic P compounds (Minear 1972, Mat-

suda & Maruyama 1985), one could expect that a considerable part of enzymatically regenerated P in meso- and eutrophic lakes is of dDNA origin.

The main aim of the study was to estimate the dDNA pool in surface waters of different meso- and eutrophic lakes and to discuss the role of various microplankton constituents in the process of dDNA liberation and decomposition. Moreover, we tried to prove that dissolved DNA represents a quantitatively significant reservoir of P for planktonic microorganisms, and that enzymatic liberation of P from this source by DNase and subsequently by 5'-PN can be one of the most effective ways of P regeneration in lakes.

METHODS

Study area and sampling. The investigations were carried out during summer stratification periods from 1991 to 1994 in 14 Polish lakes (Mazurian Lake District, Northern Poland) and 7 lakes located in Southern Germany (Table 1). The trophic status of the studied lakes varied from mesotrophy (Lake Kuc) to advanced eutrophy (Lake Tałtowisko). Samples (3 to 6 l) were collected under non-sterile conditions into polypropylene bottles from the surface layer (1 m) of the pelagic zone

Table 1. Basic morphological and limnological parameters and the mean dissolved DNA (dDNA) concentrations in the studied lakes. Lakes are arranged according to increasing trophic status. PT: total phosphorus; Chl *a*: chlorophyll *a*; TSI: trophic status index calculated from chl *a* ($TSI_{chl\ a}$) and from PT (TSI_{PT})

Lake	Area (ha)	Depth (m)		Chl <i>a</i> ($\mu\text{g l}^{-1}$)	PT ($\mu\text{g l}^{-1}$)	dDNA ^a ($\mu\text{g l}^{-1}$)	$TSI_{chl\ a}$	TSI_{PT}
		Max.	Mean					
Polish lakes								
Kuc	99	28.0	8.0	1.7	25.0	19.3	51	35
Majcz	45	16.5	3.2	1.8	31.0	27.3	54	36
Łabap	350	13.4	8.5	2.2	54.0	22.0	62	38
Kisajno	1896	25.0	8.4	3.5	67.0	16.8	65	43
Dargin	2680	37.6	10.6	3.8	69.0	17.2	65	44
Śniardwy	11340	23.4	5.8	8.2	54.0	43.1	62	51
Bełdany	941	46.0	10.0	9.9	73.0	49.1	66	53
Głębokie	47	34.0	15.0	10.7	89.0	22.6	69	54
Ryńskie	676	51.0	13.8	26.5	122.0	42.6	73	63
Jagodne	943	37.4	8.7	30.5	132.0	57.0	75	64
Mikołajskie	498	25.9	11.2	35.1	63.0	52.0	64	65
Tały	1160	44.7	13.5	40.0	77.0	43.6	67	66
Niegocin	2600	39.7	9.9	42.7	173.0	50.5	78	67
Tałtowisko	327	39.5	14.0	55.8	194.0	59.3	80	70
German lakes								
Constance	54000	254.0	90.0	4.6	28.0	13.4	52	45
Illmensee	70	–	–	11.0	31.0	38.6	54	54
Dagersee	33	12.5	6.2	14.3	32.5	25.9	54	56
Schleinsee	15	10.0	5.6	24.0	121.0	43.1	61	62
Schreckensee	30	11.5	6.4	25.8	51.3	42.4	67	63
Lengenweiler	–	–	–	37.0	112.0	41.7	72	66
Althausen Weih	14	2.8	1.3	88.0	210.0	35.4	74	68

^aMean of 2 or 3 determinations (Polish lakes), 16 determinations (Lake Constance) and 3 or 4 determinations (other German lakes)

of the studied lakes. In addition, similar samples were taken from various depths of Lake Constance and Lake Schleinsee to study the dDNA depth profiles.

dDNA determination. Lake water samples were filtered through 0.2 μm polycarbonate membrane filters (Nuclepore) within 1 to 3 h after sampling and divided into four 0.1 to 1.0 l subsamples. Three of them were replicates, the fourth one was supplemented with DNA from calf thymus (Serva, Cat. No. A18560) to 50–100 $\mu\text{g l}^{-1}$ final concentration and treated in parallel to replicates for internal standardization of the measurement.

dDNA was precipitated by slow addition of CTAB (cetyltrimethyl-ammonium bromide, Sigma Chemical Co.) stock solution (2 ml for each 100 ml of sample) to stirred filtrates. Filtrates treated with CTAB were frozen at -28°C and stored (no longer than 2 wk) for further analysis. The CTAB stock solution contained 5 g CTAB in 1000 ml of 0.5 M NaCl. Immediately before the dDNA assay, frozen subsamples were thawed at room temperature and the insoluble DNA-CTA precipitate was collected by centrifugation ($3400 \times g$, 30 min). The DNA-CTA complex was converted into soluble DNA- Na_2 salts by solubilizing the precipitate in 10 ml of 1 M NaCl (3 h at 40°C). After solubilization, the dissolved DNA was separated from insoluble contaminants by additional centrifugation ($3000 \times g$, 15 min). Finally, we added 0.1 ml of 1 M Tris-Na buffer (pH 8.3) and 0.1 ml ($10 \mu\text{g ml}^{-1}$) of DAPI (di-amidino-phenyl-indole, Serva GmbH) to 2.3 ml of clear supernatant solution and after 10 min incubation in the darkness the fluorescence of the DNA-DAPI complex was measured with a Perkin Elmer Spectro-Fluorimeter LS 50 B or Shimadzu RF 1501 at 365 nm (excitation) and 445 nm (emission). After subtraction of the auto-fluorescence of DAPI free samples (DeFlaun et al. 1986), dDNA concentrations were calculated from a standard curve, taking precipitation efficiency into consideration (Karl & Bailiff 1989).

Stock solutions (1 mg ml^{-1}) of DAPI and DNA from calf thymus were prepared in deionized water and stored at -28°C for no longer than 1 mo. They were thawed at room temperature and diluted with deionized water to required concentrations just before the DNA assay. All DNA solutions for standard curves were prepared in 1 N NaCl. Further details of the dDNA assay are given by Siuda & Güde (1996a).

DNase activity. Triplicate water samples (19.5 ml) were supplemented with 0.2 ml of calf thymus DNA solution in deionized water (30 to 600 $\mu\text{g DNA l}^{-1}$). Sodium azide solution (0.3 ml; final concentration 0.3%) was added to prevent bacterial growth. At time 0 and after 24 to 48 h of incubation, 10 ml subsamples were filtered through 0.2 μm Nuclepore filters. DNA was determined directly (without precipitation) in all

filtrates by the method described above. Hydrolysis rates (v) for each DNA concentration were expressed as a decrease of DNA concentration in filtrates per liter and hour. The maximal ambient dDNA hydrolysis rate (v_{max}) for each analysed water sample was calculated from DNA hydrolysis kinetics curves (see 'Results' and Fig. 5b) using the non-linear regression analysis software 'Enzfitter' (Elsevier Biosoft, UK).

Other analyses. Algal primary production and extracellular release of photosynthetic products were measured with the ^{14}C method (Siuda et al. 1991). Bacterial secondary production was determined by the ^3H -thymidine method according to Chróst et al. (1988). Bacteria were counted directly in epifluorescence microscope after staining (5 min) with di-amidino-phenyl-indole (DAPI) at $2.7 \times 10^{-3} \mu\text{M}$ final concentration (Güde et al. 1985). Algal pigments (chl *a* and pheophytin) were determined spectrophotometrically, after extraction with 90% acetone (Golterman & Clymo 1969) in German lakes or with 96% ethanol according to Marker et al. (1980) in Polish lakes. Total and dissolved organic phosphorus was measured according to Koroleff (1983).

RESULTS

Concentrations of dDNA in lake water

dDNA concentration in the surface layer of all studied lakes varied between 0.5 and 88 $\mu\text{g l}^{-1}$ during the summer stratification period. Usually, the smallest amounts of dDNA (up to 20 $\mu\text{g l}^{-1}$) were found in mesotrophic lakes (Dargin, Kisajno, Kuc, Constance and Dagersee). Medium (20 to 40 $\mu\text{g l}^{-1}$) and highest (40 to 60 $\mu\text{g l}^{-1}$) dDNA contents were observed in eutrophic lakes (Table 1).

Calculations made for all the obtained results (Table 2) showed that the mean size of the dDNA pool correlated well with the trophic status index of lakes (TSI_{PT} : $r = 0.62$, $n = 21$, $p < 0.01$; $\text{TSI}_{\text{chl } a}$: $r = 0.79$, $n = 21$, $p < 0.01$) as well as with the Secchi disc visibility ($r = 0.74$, $n = 24$, $p < 0.01$). Only between dDNA and basic algal parameters (i.e. chl *a* + pheophytin concentration and total primary production) were positive and statistically significant correlations found ($r = 0.74$, $n = 21$, $p < 0.01$ and $r = 0.78$, $n = 22$, $p < 0.01$, respectively). No statistically significant correlation of dDNA with bacterial number and secondary production was observed (Table 2; $r = 0.32$, $n = 35$ and $r = 0.39$, $n = 22$, respectively).

Changes of dDNA concentrations and basic biological parameters (chl *a*, bacterial number and DNase activity) were examined more extensively in mesotrophic Lake Constance and eutrophic Lake Schlein-

Table 2. Correlation of dDNA concentrations with basic limnological and biological parameters in surface water of the studied lakes. Correlations and regressions (\pm SD) were calculated from individual pairs of the data collected during summer periods 1991 to 1994. ns: nonsignificant

Correlated parameter	r	n	p	Linear regression	
				Slope	Constant
Log (Secchi disc visibility)	0.74	24	<0.01	-40.4418 \pm 7.901	48.3281 \pm 11.9354
TSP _{PT} ^a	0.62	21	<0.01	1.0191 \pm 0.2945	-29.9179 \pm 11.3154
TSP _{chl a} ^b	0.79	21	<0.01	17.6587 \pm 8.9469	0.9748 \pm 0.1764
Log(chl a + pheo.)	0.74	21	<0.01	27.5063 \pm 4.0245	1.1656 \pm 0.5514
Log(total primary production) ^b	0.78	22	<0.01	0.5032 \pm 0.0900	15.9310 \pm 11.5087
Bacterial number	0.32	35	ns	-	-
Bacterial secondary production	0.39	22	ns	-	-

^aCalculated according to Carlson (1977)
^bIncluding cellular production, extracellular release and bacterial uptake of released products

see. In surface waters (1 m depth) of Lake Constance (Fig. 1) the lowest amounts of dDNA (below $5 \mu\text{g l}^{-1}$) were observed at the beginning of June during the clear water phase (chl *a* conc. 1.3 to $5.7 \mu\text{g l}^{-1}$). In July and in the first half of August, dDNA concentrations were nearly stable (10 to $15 \mu\text{g l}^{-1}$) and reached a maximum ($22 \mu\text{g l}^{-1}$) after the first phytoplankton bloom (chl *a* conc. $10.3 \mu\text{g l}^{-1}$). Bacterial numbers were relatively high in early summer (7.7 to 23.6×10^6 cells ml^{-1}) and decreased twice (3.2 to 9.4×10^6 cells ml^{-1}) in the second part of our investigation period.

Both dDNA depth profiles obtained for Lake Constance (Fig. 2) were similar and exhibited the greatest

variations and maxima at 2, 10, 20 and 30 m depth (17 to $19 \mu\text{g l}^{-1}$) in November and at 2 and 4 m depth (9.0 to $11.3 \mu\text{g l}^{-1}$) in December. dDNA concentrations were relatively constant in the deeper parts of the lake with minimal values below 150 m depth (11.9 and $3.4 \mu\text{g l}^{-1}$, respectively).

In Lake Schleinsee we found 2 dDNA maxima (Fig. 3). The first in the epilimnion (1 to 2 m, 47 to $58 \mu\text{g l}^{-1}$) and the second (55.6 to $60.8 \mu\text{g l}^{-1}$) in the anoxic, H_2S rich zone above the bottom sediments (1 to 1.5 m from the bottom). The dDNA concentration was negatively correlated with bacterial numbers ($r = -0.87$, $n = 5$, nonsignificant) in the surface waters (0 to 4 m), but positively in the profundal zone (5 to 10 m) ($r = 0.94$, $n = 5$, $p < 0.05$).

Our laboratory and field observations of dDNA degradation by lake microorganisms showed that bacteria can effectively utilize calf thymus

DNA added to lake water samples (Fig. 4). The kinetic curves of enzymatic dDNA hydrolysis (Fig. 5) could rarely (e.g. Lake Constance, 15 June) be described by Michaelis-Menten equations. They often consisted of 2 steps (e.g. Lake Constance, 6 July) and sometimes (particularly in highly eutrophic lakes) were S-shaped (e.g. Lake Schleinsee, 14 July). Therefore, precise calculations of K_m was difficult. However, to compare the hydrolytic potential for DNA decomposition in various lake waters or microplankton size fractions, we tried to estimate experimentally v_{max} , the maximal ambient dDNA hydrolysis rate, directly related neither to V_{max} nor to v from Michaelis-

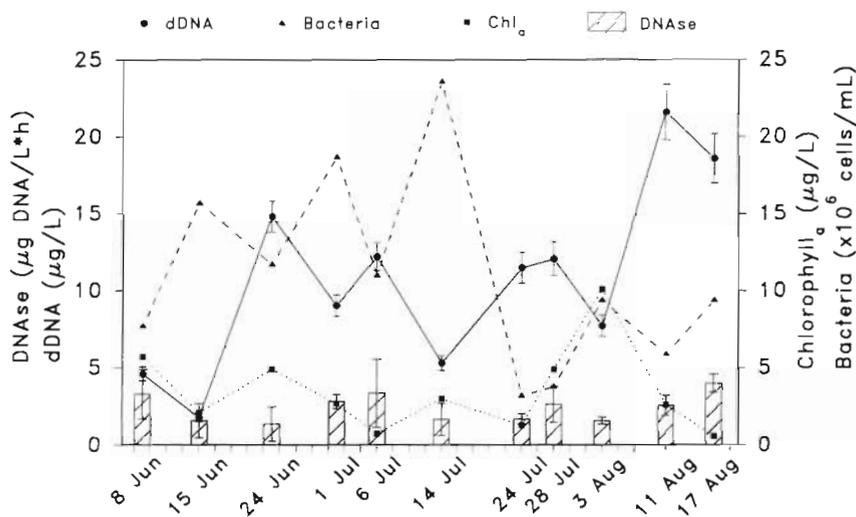


Fig. 1. Changes in dDNA concentration, bacterial number, chl *a* and DNase activity (v_{max}) in surface waters of Lake Constance in summer 1992. Each value of dDNA concentration and DNase activity represents a mean of triplicate determinations \pm standard error

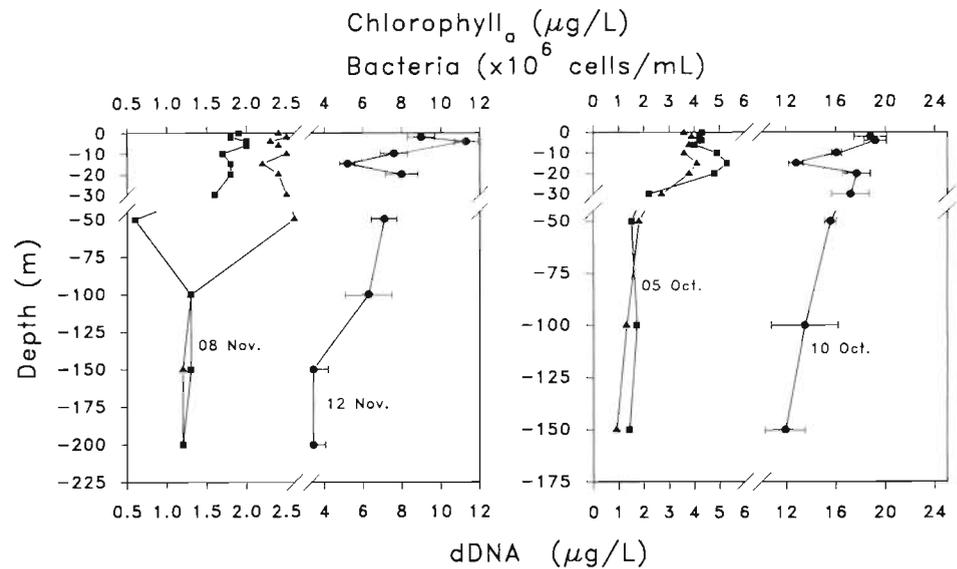


Fig. 2. Vertical distribution of dDNA, chl *a* and bacteria in Lake Constance. (●) mean of triplicate determinations of dDNA \pm SD, (▲) bacterial number, (■) chl *a* concentration

Menten equations. For this rough approximation we assumed that in the case of 2 step kinetic curves only the first 'step' describes to some extent dDNA hydrolysis at ambient dDNA concentrations. The analysis of the relationship between the dDNA hydrolysis rates and dDNA concentrations much higher than those observed in lake water samples has no ecological sense and may be neglected.

The ability of natural DNAses to degrade calf thymus DNA (the maximal ambient dDNA hydrolysis rate, v_{max}) did not exhibit larger changes (1.4 to 4.0 $\mu\text{g DNA l}^{-1} \text{h}^{-1}$) in the epilimnion of Lake Constance and fluctuated around 2.5 $\mu\text{g DNA l}^{-1} \text{h}^{-1}$ (Fig. 1). It was correlated neither with bacterial numbers nor with concentration of algal pigments.

The participation of free and cell bound DNase to the total hydrolytic potential of the DNase in lake water calculated for selected German and Polish lakes varied drastically from 3 to 97% (Fig. 6). Cell bound DNase activity was mainly coupled to the picoplankton (0.2 to 1.0 μm) size fraction (64%) whereas 33.9% of the total cell bound DNase activity was found in the fraction $>1.0 \mu\text{m}$.

DISCUSSION

Our results showed a ubiquitous distribution of dDNA in lake waters. Although the dissolved deoxyribonucleic acid concentrations in surface waters varied

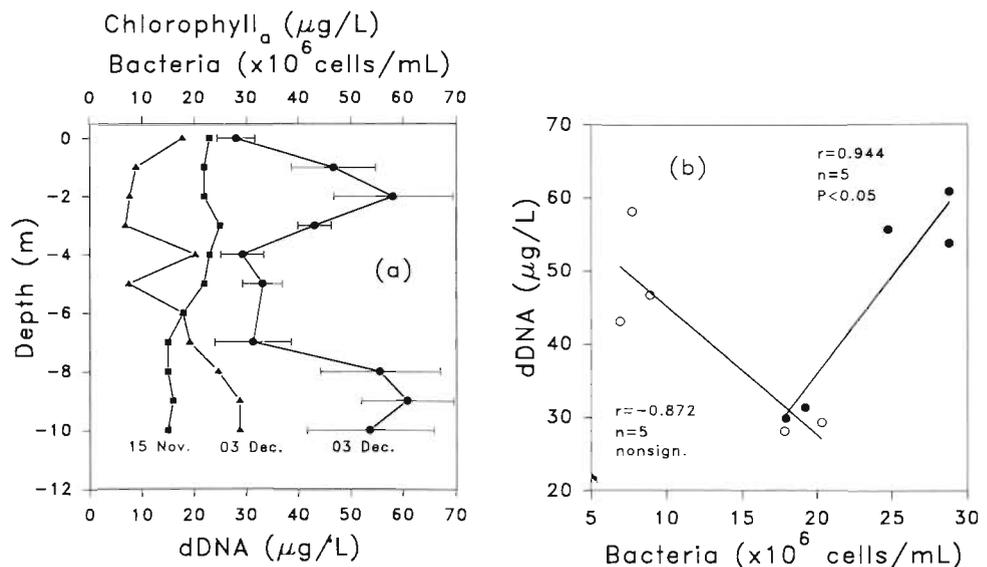


Fig. 3. (a) Vertical distribution of dDNA (●, mean of triplicate determinations \pm SD), chl *a* (■) and the bacteria (▲) in Lake Schleinsee; (b) correlation between dDNA concentration and bacterial numbers in epilimnetic (0 to 4 m, ○) and profundal (6 to 10 m, ●) waters of the lake

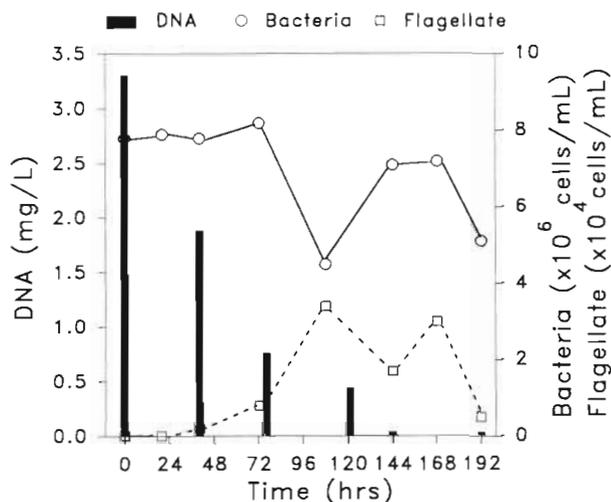


Fig. 4. DNA utilization by natural bacterial populations from Lake Constance. Surface (1 m) water sample was prefiltered through a 1.0 μ m Nuclepore membrane filter, enriched with 3 mg of calf thymus DNA and incubated at room temperature. Flagellates in the culture developed spontaneously

among the studied lakes, with values ranging from below the detection limit ($0.5 \mu\text{g l}^{-1}$) up to $88 \mu\text{g l}^{-1}$, the mean dDNA content did not usually exceed $20 \mu\text{g l}^{-1}$ in mesotrophic and $60 \mu\text{g l}^{-1}$ in eutrophic environments (Table 1). Since the total organic P content varied from 25 to $50 \mu\text{g l}^{-1}$ and the range of DOP concentrations commonly changed between 5 and $15 \mu\text{g P l}^{-1}$ in these lakes, one could estimate (presuming that DNA-P amounts for about 8% of total DNA weight) that dDNA phosphorus may contribute approximately 5 to 60% of DOP and 0.2 to 8.0% of the total P pool.

The dDNA pool was generally proportional to the Secchi disc visibility ($r = 0.74$, $n = 24$, $p < 0.01$). Therefore, one could expect also a positive correlation between dDNA and microplankton biomass as indicated by algal pigments and bacterial numbers. The role of phytoplankton in dDNA liberation processes was postulated by Minear (1972), who found considerable amounts of dDNA in pure algal cultures and a highly significant correlation of dDNA concentrations with algal pigments in the environment. Reisser et al. (1993) reported both viral and algal DNA release in *Chlorella*-virus cultures. Also, Beebee (1991) in his paper concluded that the bulk of dDNA in eutrophic lakes in England exists in linear rather than in circular molecules and is within the size range of eukaryotic exons. However, Paul & Carlson (1984), Paul et al. (1987) and DeFlaun et al. (1986) questioned the significant contribution of algal DNA to the total dDNA pool and pointed out the role of bacteria in supplementing dDNA. Similarly, Turk et al. (1992) demonstrated strong evidence for the production of substantial

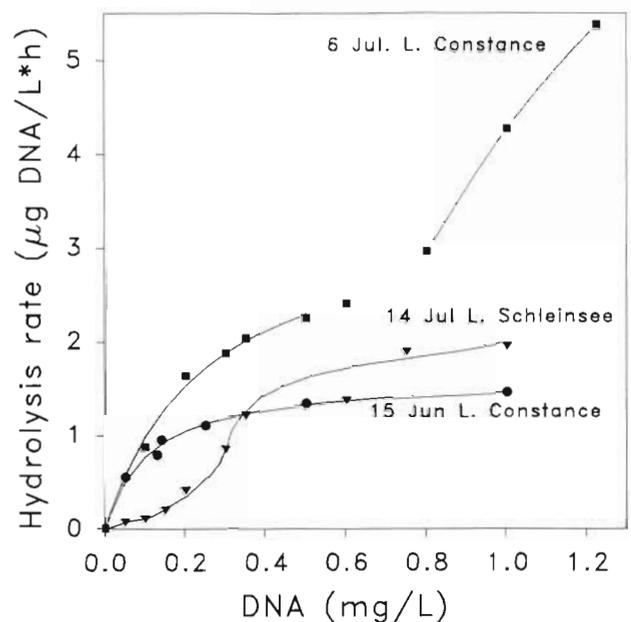


Fig. 5. Selected examples of kinetics of DNA hydrolysis by DNases in various lake water samples

amounts of dDNA by nanoflagellate grazing on bacteria in a continuous seawater culture. Although our observations confirmed the role of phytoplankton in supplementing the dDNA pool, we could not confirm the liberation of free DNA by bacteria (Table 1, Fig. 3). The only exception was the profundal zone of the eutrophic Lake Schleinsee (Fig. 3) where a peak of dDNA concentrations (54 to $62 \mu\text{g l}^{-1}$) was correlated with relatively high bacterial numbers (2.5 to 2.8×10^7 cells ml^{-1}). However, we speculate that in this case accumulation of dDNA in profundal waters of Lake Schleinsee could rather be an effect of intensified sedimentation of DNA rich detritus from the upper zone of the lake or/and inhibition of DNase activity by H_2S . Hoppe (1986) recognized H_2S as a noncompetitive inhibitor of several bacterial ectoenzymes (i.e. α - and β -glucosidase, N-acetyl-glucosaminidase, alkaline phosphatase).

Due to methodological difficulties, all results of DNase activity measurements should be treated with great caution and may be only indirectly related to conditions of the natural environment. However, they permit one to draw some general conclusions concerning P regeneration processes in a lake. Firstly, free DNA is quantitatively important as a potentially available P reservoir for lake microplankton. Secondly, DNase activity in surface waters of Lake Constance was mainly extracellular or (similarly to 5'-nucleotidase activity) coupled to the plankton size fraction 0.2 to $1.0 \mu\text{m}$, which confirms our earlier assumption that only bacteria can use dDNA as a P source (Siuda &

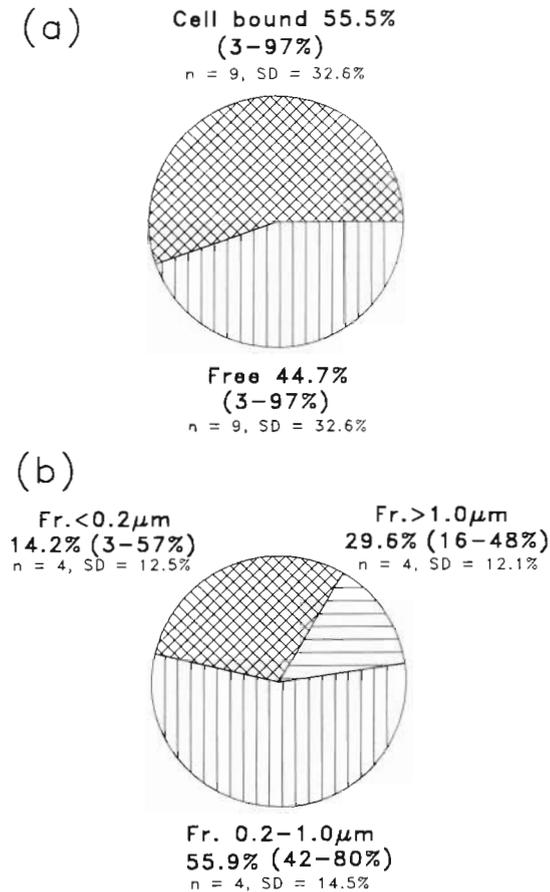


Fig. 6. (a) Mean relative contribution of free and cell bound DNase in the lake water and (b) participation of various plankton size fractions to the total DNase activity (v_{DNase}) in surface waters of Polish and German lakes. Ranges of values in parentheses

Güde 1996b). Thirdly, DNase activity was relatively stable during the whole investigation period, which probably ensured constant supplementation of surface water with nucleotides. Thus, in comparison to non-continuous phosphate liberation from qualitatively and quantitatively poorly defined phosphomonoesters by alkaline phosphatase (Taft et al. 1977, Solorzano 1978, Chróst et al. 1986, Siuda & Güde 1994a), the more permanent regeneration of P from dDNA (via nucleotides and 5'-nucleotidase), although probably less efficient (Siuda & Güde 1996b), may be regarded as one of the most important factors that favors bacterial populations in competition with algae for regenerated P.

Summarizing the observations made in the present study, we have found that concentrations of dDNA were related to the trophic status of the investigated lakes. Moreover, they were positively correlated with tested algal parameters (algal pigments, primary production) and did not fit to any bacterial determinants (bacterial number, secondary production, DNase

activity). Therefore we can conclude that, in surface waters of eutrophic (and perhaps mesotrophic) lakes during the summer stratification period, eukaryotic organisms (algae, zooplankton) are a source of dDNA whereas bacteria tend to utilize the dDNA pool.

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