

# Effects of *Tubifex tubifex* (Oligochaeta: Tubificidae) on N-mineralization in freshwater sediments, measured with $^{15}\text{N}$ isotopes

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**ABSTRACT:** Sediment cores containing different densities of *Tubifex tubifex*, ranging from 0 to 70 000 ind.  $\text{m}^{-2}$ , were incubated in the laboratory. Rates of  $\text{O}_2$  and  $\text{NO}_3^-$  uptake,  $\text{NH}_4^+$  production, nitrification and denitrification were determined from sediment-water fluxes. Pore water  $\text{NH}_4^+$  was measured at the end of the experiment. At natural densities,  $\sim 50\,000$  ind.  $\text{m}^{-2}$ , there were increased rates of  $\text{O}_2$  consumption ( $\times 2$ ), denitrification of water phase  $\text{NO}_3^-$  ( $\times 3$ ) and  $\text{NH}_4^+$  efflux ( $\times 26$ ). Nitrification was stimulated at low worm densities, but inhibited at higher worm densities. The transport of reduced compounds and organic matter, with the fecal pellets, to the sediment surface stimulated anoxic conditions in the inhabited microcosms. These anoxic conditions led to increased rates of denitrification and were responsible for the decrease in nitrification at higher worm densities. Approximately 25% of the  $\text{NO}_3^-$  produced by nitrification within the sediment was subsequently denitrified. Denitrification was responsible for 25% of the  $\text{NO}_3^-$  disappearance from the system. The higher rates of denitrification were counterbalanced by higher rates of  $\text{NH}_4^+$  flux from the sediment. It is likely, however, that the presence of *T. tubifex* resulted in a net loss of nitrogen that could otherwise have been used by the primary producers.

**KEY WORDS:** Bioturbation · Nitrification · Denitrification · Sediments

## INTRODUCTION

Macrobenthos, through their burrowing, feeding, locomotive, respiratory and excretory activities, play an important role in mediating both physical and chemical processes near the sediment-water interface (Fisher 1982, McCall & Tevesz 1982). Tubificid worms are often, along with chironomid larvae, the most abundant macrofauna species in eutrophied stream and lake sediments. Population densities can be as high as millions of individuals per  $\text{m}^2$  (Palmer 1968). Many studies concerning the effect of these worms on the physical properties of the sediment have been carried out (McCall & Fisher 1980, Fisher 1982, McCall &

Tevesz 1982). However, their role in nutrient cycling, specifically nitrogen cycling, has received less attention (Kikuchi & Kurihara 1977, 1982, Chaterpaul et al. 1979, 1980).

Tubificid worms feed primarily in the top 2 to 8 cm of sediment (McCall & Tevesz 1982), adjusting their feeding depth to lower strata at higher worm densities (Robbins et al. 1979, McCall & Fisher 1980). Tubificid worms are 'conveyor belt' feeders (Rhoads 1974). They typically live and feed head down in the sediment. Some portion of the posterior of the worms may project above the sediment-water interface. The worms selectively ingest silt and clay particles at depth and digest the attached microflora, primarily bacteria (Davis 1974). Fecal pellets are deposited at the sediment-water interface, where they may form a pelletized layer.

Studies of the effect of bioturbation by marine infauna on nitrification and denitrification processes

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(Kristensen et al. 1991, Pelegrí et al. 1994, Pelegrí & Blackburn 1995) have shown that the particular feeding and burrowing strategies of the various animals can lead to variable stimulation of aerobic respiration, nitrification and denitrification. The aim of this study is to clarify the effect of tubificid worms on nitrogen metabolism and on fluxes across the sediment-water interface.

## MATERIAL AND METHODS

Undisturbed sediment cores (19.5 cm long and 3.5 cm inner diameter), containing no visible macrofauna, were collected in Giber Å stream (Denmark). This stream is dominated by runoff from agricultural areas and receives fluctuating amounts of  $\text{NO}_3^-$  and organic matter (Nielsen et al. 1990). *Tubifex tubifex* Müller (Oligochaeta: Tubificidae), collected from sieved (500  $\mu\text{m}$ ) Universitetsparken Lake sediment, were added to these cores ( $n = 11$ ) at densities ranging from 0 to 70 ind. core $^{-1}$  (70 000 ind. m $^{-2}$ ). The worms dug immediately into the sediment. The cores were placed in a reservoir containing well-aerated tap water, and kept in darkness at  $10.8 \pm 0.8^\circ\text{C}$ . An atmospheric  $\text{NH}_4^+$  trap, consisting of 3 gas washing bottles containing 0.1 M  $\text{H}_2\text{SO}_4$ , 50 mM  $\text{K}_2\text{HPO}_4$  and distilled water connected in line, was installed in the aerating circuit between the air pump outlet and the reservoir. Ammonium concentrations in the reservoir water ranged from 0.3 to 1.1  $\mu\text{M}$  during the experiment.

**Experimental procedure.** Every second day  $\text{O}_2$ ,  $\text{NH}_4^+$  and  $\text{NO}_3^- + \text{NO}_2^-$  fluxes were measured in order to determine when steady state was reached. This was done by closing the cores for 2 h, after which samples from the overlying water were collected and immediately frozen. Eleven days after setup, when steady state was reached,  $^{15}\text{NO}_3^-$  (5 mM  $\text{K}^{15}\text{NO}_3^-$ , 99%  $^{15}\text{N}$  stock solution) was added to the reservoir water to a final concentration of 38.3  $\mu\text{M}$   $\text{NO}_3^-$  (69.7%  $^{15}\text{NO}_3^-$ ). After 2 d, the cores were stoppered and incubated for 4 to 5 h, which resulted in less than 20%  $\text{O}_2$  depletion. At the end of the incubation the stoppers were removed and water samples were taken and stored. Samples for analysis of dissolved  $\text{N}_2$  gas were stored in 6 ml glass vials (Exetainer, Labco Ltd, Buckinghamshire, UK), containing 2%  $\text{ZnCl}_2$  in order to stop microbial activity. Nitrogen gas was extracted from these water samples with 0.9 ml helium, and analyzed by mass spectrometry. Samples for analyzing  $\text{NH}_4^+$ ,  $\text{NO}_3^- + \text{NO}_2^-$  and % $^{15}\text{NO}_3^-$  enrichment were collected in plastic vials and immediately frozen.

In order to test the applicability of the denitrification assay in our bioturbated sediment, the water in the reservoir was exchanged with water containing 73  $\mu\text{M}$   $\text{NO}_3^-$  (83%  $^{15}\text{NO}_3^-$ ) and the cores were incubated

again following the same procedure as described above. Two days later a final incubation took place (reservoir water containing 62.1  $\mu\text{M}$   $\text{NO}_3^-$ , 81%  $^{15}\text{NO}_3^-$ ).

At the end of the experiment the cores were cut into 0–0.5, 0.5–1, 1–2, 2–3 and 3–5 cm sections. Pore water was obtained by centrifuging the sediment for 10 min, at 3000 rpm ( $2000 \times g$ ) in double centrifuge tubes (Blackburn 1980). The supernatant was immediately collected and frozen for later analysis. Unfortunately, due to the small volumes of pore water, the analyses were run on single samples.

**Analysis.** Oxygen was measured directly in the microcosms with an oxygen microsensor provided with a guard cathode (Revsbech et al. 1989).  $\text{NO}_3^- + \text{NO}_2^-$  was measured by standard methods (Grasshoff et al. 1983) in a flow injection analyzer (Tecator, Höganäs, Sweden) and analyses of  $\text{NH}_4^+$  were done by the salicylate hypochlorite method (Bower & Holm-Hansen 1980). The formation of  $^{15}\text{N}$  labeled dinitrogen pairs ( $^{14}\text{N}^{15}\text{N}$  and  $^{15}\text{N}^{15}\text{N}$ ) by denitrification was measured on an isotope ratio mass spectrometer as described by Nielsen (1992). The concentration of the dinitrogen pairs in the samples was calculated by multiplying their ratio ( $^{14}\text{N}^{15}\text{N}$  or  $^{15}\text{N}^{15}\text{N}/\text{total N}$ ) by the  $\text{N}_2$  concentration in the water (Weiss 1970). The % $^{15}\text{NO}_3^-$  enrichment was analyzed on a mass spectrometer after using denitrifying bacteria cultures to transform  $\text{NO}_3^-$  to gaseous  $\text{N}_2$  (Risgaard-Petersen et al. 1993).

**Calculations.** Rates of denitrification were measured using the isotope pairing technique (Nielsen 1992). The formation rates of single- ( $^{14}\text{N}^{15}\text{N}$ ) and double-labeled ( $^{15}\text{N}^{15}\text{N}$ ) dinitrogen pairs were used to calculate the denitrification of  $\text{NO}_3^-$  coming from the overlying water ( $dw$ ) and of  $\text{NO}_3^-$  generated within the sediment by nitrification ( $dn$ ):

$$d_{15} = ({}^{14}\text{N}^{15}\text{N}) + 2({}^{15}\text{N}^{15}\text{N}) \quad (\text{Koike \& Hattori 1978});$$

$$d_{14} = d_{15} \frac{({}^{14}\text{N}^{15}\text{N})}{2({}^{15}\text{N}^{15}\text{N})},$$

$$dw = d_{15} \frac{100}{e_{15}},$$

$$dn = d_{14} + d_{15} - dw \quad (\text{Nielsen 1992})$$

where  $e_{15}$  is the % $^{15}\text{NO}_3^-$  enrichment of the reservoir water.

Nitrification was calculated by a modification of the equation described by Blackburn (1993) for  $^{15}\text{NH}_4^+$  turnover in sediments:

$$d = (d - i) \frac{\ln(e_0 - 0.003663) - \ln(e_t - 0.003663)}{\ln \frac{c_t}{c_0}} + dn$$

where  $d$  is the nitrification rate,  $i$  is the  $\text{NO}_3^-$  consumption rate,  $e$  is the % $^{15}\text{NO}_3^-$  enrichment,  $c$  is the  $\text{NO}_3^-$

concentration in the water phase ( $0 = \text{initial}$ ,  $t = \text{final}$ ) and 0.003663 is the natural abundance of  $^{15}\text{N}$  in nature.

The rates of  $dw$  and  $\text{NO}_3^-$  uptake are dependent on the  $\text{NO}_3^-$  concentration in the water column, which was different in the various incubations. These rates were normalized to  $50 \mu\text{M}$   $\text{NO}_3^-$  concentration in the overlying water for all incubations.

## RESULTS

Oxygen consumption (Fig. 1A),  $\text{NH}_4^+$  fluxes (Fig. 1B) and  $\text{NO}_3^-$  consumption (Fig. 1C) were stimulated by the presence of the worms.

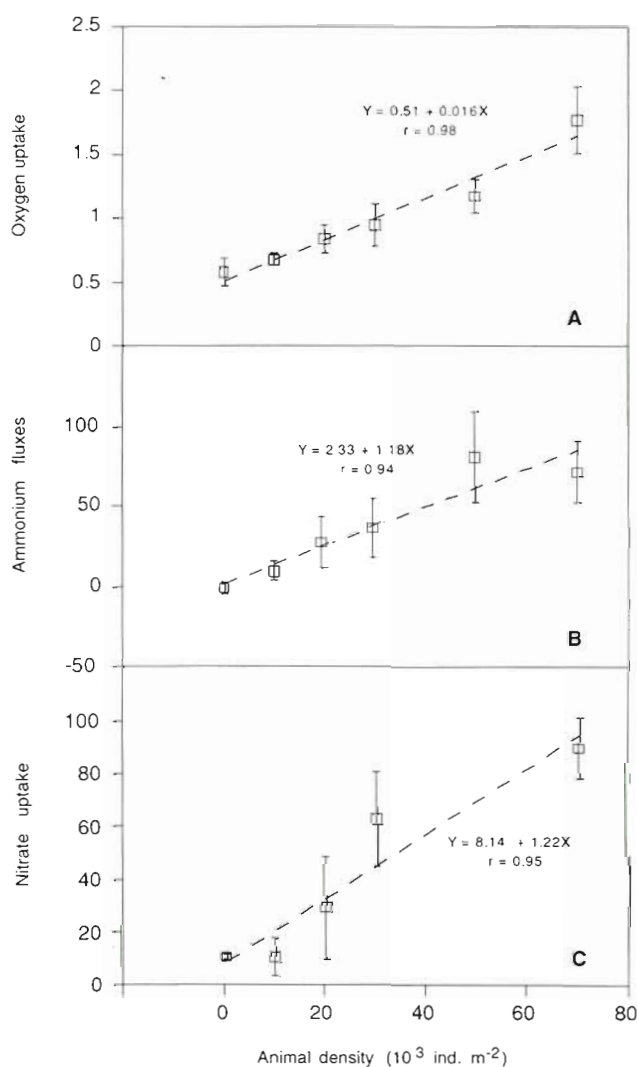


Fig. 1. *Tubifex tubifex*. (A) Oxygen uptake ( $\text{mmol m}^{-2} \text{ h}^{-1}$ ), (B) ammonium effluxes ( $\mu\text{mol m}^{-2} \text{ h}^{-1}$ ) and (C) nitrate uptake ( $\mu\text{mol m}^{-2} \text{ h}^{-1}$ ) corrected to  $50 \mu\text{M}$   $\text{NO}_3^-$  in the water phase, versus tubificid worm density. Each data point is the average of 6 replicates, except for microcosms containing  $10000 \text{ ind. m}^{-2}$  with only 3 replicates. Error bars represent standard errors

The isotope pairing technique assumes that the increase in concentration in the water by the addition of  $^{15}\text{NO}_3^-$  does not influence denitrification of the sediment-produced  $^{14}\text{NO}_3^-$  and that the  $\text{NO}_3^-$  species ( $^{15}\text{NO}_3^-$  and  $^{14}\text{NO}_3^-$ ) are uniformly mixed in the denitrification zone. Since the production of unlabeled  $\text{N}_2$  ( $^{14}\text{N}^{14}\text{N}$ ) is not measured directly but estimated from the production of single- ( $^{14}\text{N}^{15}\text{N}$ ) and double-labeled ( $^{15}\text{N}^{15}\text{N}$ ) dinitrogen pairs, changes in the ratio of  $dw$  (denitrification of  $\text{NO}_3^-$  coming from the overlying water) to  $dn$  (denitrification of  $\text{NO}_3^-$  generated within the sediment oxic layer) due to bioturbation may underestimate the  $dn$  values. This can be tested by incubating with different concentrations of  $^{15}\text{NO}_3^-$ . Thus at higher  $^{15}\text{NO}_3^-$  levels more  $^{14}\text{NO}_3^-$  will be paired with  $^{15}\text{NO}_3^-$  to form measurable  $^{14}\text{N}^{15}\text{N}$  and less  $^{14}\text{N}^{14}\text{N}$  is formed. Rates of  $dn$ , measured with  $\text{NO}_3^-$  concentrations in the overlying water ranging from  $38.3$  to  $73 \mu\text{M}$ , were not significantly different ( $p < 0.001$ ). Thus the addition of  $38.3 \mu\text{M}$   $\text{NO}_3^-$  to the overlying water was sufficient to avoid underestimation of  $dn$  rates. Rates of nitrification and  $dn$  (Fig. 2A) increased significantly ( $p < 0.05$ ) by increasing the worm density from  $0$  to  $20000 \text{ ind. m}^{-2}$ . However, nitrification rates decreased significantly ( $p < 0.05$ ) when worm density increased from  $20000$  to  $70000 \text{ ind. m}^{-2}$ . Around 25% of the total  $\text{NH}_4^+$  nitrified was subsequently denitrified in the suboxic sediment layer.

Denitrification of  $\text{NO}_3^-$  coming from the overlying water ( $dw$ ) was stimulated by the presence of the worms (Fig. 2B). An average of 25% of the  $\text{NO}_3^-$  uptake by the sediment was removed from the system by denitrification. Rates of  $dw$  represented from 20 to 90% of the total denitrification rates in microcosms containing worm densities ranging from  $0$  to  $70000 \text{ ind. m}^{-2}$ .

Ammonium concentrations increased with depth, but the increase, compared with the control, was less with  $< 20000 \text{ ind. m}^{-2}$  and greater with  $> 20000 \text{ ind. m}^{-2}$  (Fig. 3).

## DISCUSSION

In our experiments, oxygen uptake was increased 3-fold at a tubificid density of  $70000 \text{ ind. m}^{-2}$ , whereas McCall & Fisher (1980) reported a 2-fold increase at  $100000 \text{ ind. m}^{-2}$ , 20% of which was due to worm respiration. Our results indicated that the worms ( $20000 \text{ ind. m}^{-2}$ ) increased denitrification of  $\text{NO}_3^-$  coming from the overlying water by 175%. Chatterpaul et al. (1980) reported that *Limnodrilus hoffmeisteri* and *Tubifex tubifex* ( $12000$  to  $16000 \text{ ind. m}^{-2}$ ) increased denitrification rates by only 80%, in spite of  $\text{NO}_3^-$  concentrations

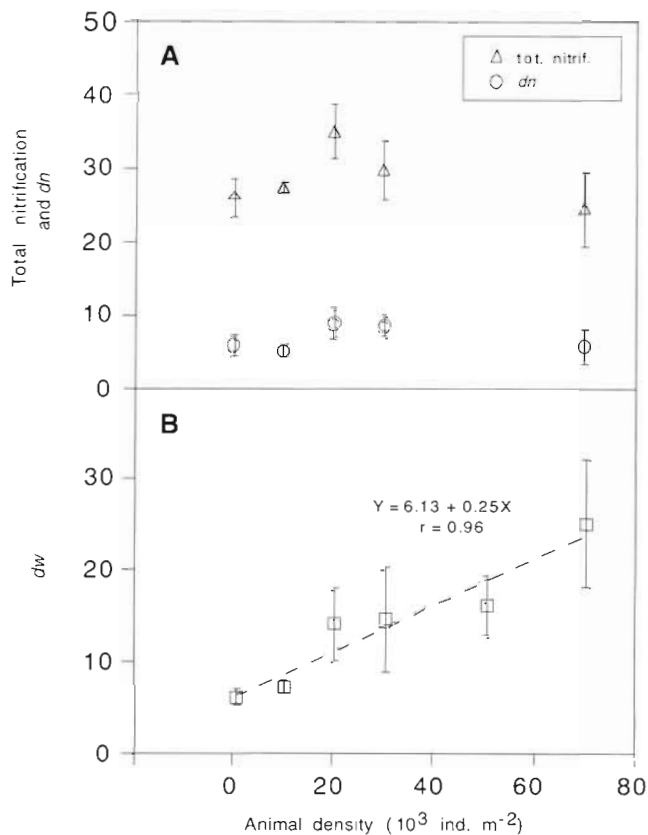


Fig. 2. *Tubifex tubifex*. (A) Total nitrification and denitrification of  $\text{NO}_3^-$  generated within the sediment ( $dn$ ) and (B) denitrification of  $\text{NO}_3^-$  coming from the overlying water ( $dw$ ), corrected to  $50 \mu\text{M}$   $\text{NO}_3^-$  in the water phase, versus tubificid worm density. Each data point is the average of 6 replicates, except for microcosms containing  $10000 \text{ ind. m}^{-2}$  with only 3 replicates. Error bars represent standard errors. Units:  $\mu\text{mol m}^{-2} \text{ h}^{-1}$

in the overlying water 4 times higher in their experiment.

The increase in oxygen consumption rates and in nitrate uptake, in bioturbated microcosms, could be associated with a change in the characteristics of the surface sediment, due to the particular feeding strategy of *Tubifex tubifex* or to active irrigation of the burrow by the worms (Pelegri et al. 1994). Tubificid worms are capable of affecting many sediment properties by the formation of a stable layer of fecal pellets at the sediment surface. Egested material is packaged into mucus-bound pellets which may maintain their integrity for some time after deposition on the sediment surface (Fisher 1979). McCall & Fisher (1980) reported increases in sediment porosity of 80% at the 0–1.5 cm layer due to fecal pellets (*Limnodrilus hoffmeisteri* and *T. tubifex*,  $3000 \text{ ind. m}^{-2}$ ). They believed that changes in sediment porosity were responsible for an increased diffusion of chloride tracer in sediment containing *T.*

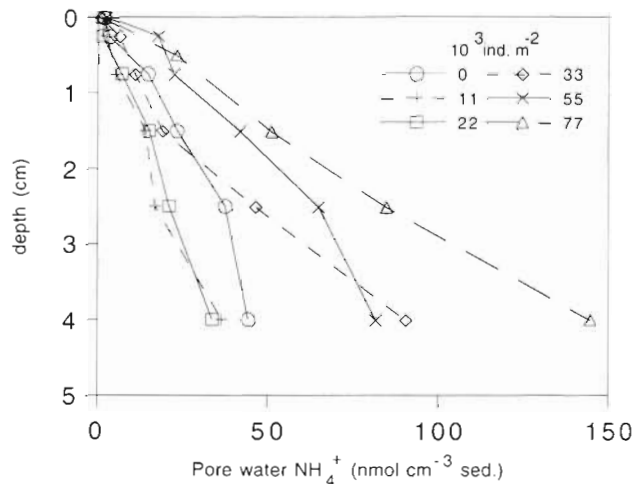


Fig. 3. *Tubifex tubifex*. Sediment concentration profiles of pore water  $\text{NH}_4^+$  in microcosms containing tubificid worm densities ranging from 0 to  $70000 \text{ ind. m}^{-2}$

*tubifex* ( $100000 \text{ m}^{-2}$ ). Wood (1975), however, attributed an increased diffusion of rhodamine B into sediment to irrigation by tubificid worms. Increased porosity and/or diffusivity results in a deeper penetration by oxygen (Davis 1974, Fukuhara 1987) and  $\text{NO}_3^-$ , but the transport of reduced material to the sediment surface and the accumulation of organically enriched fecal pellets can result in increased rates of oxygen and  $\text{NO}_3^-$  utilization. McCall & Fisher (1980) reported a 2-fold increase in organic content at the 0–1.5 cm layer, probably attributable to the higher organic content in fecal pellets compared to bulk sediment (35%; Brinkhurst et al. 1972).

The pore water  $\text{NH}_4^+$  content increased in sediments at worm densities  $>20000 \text{ ind. m}^{-2}$ . Similar increases occurred in the 0–5 cm layer of rice-field soil containing *Limnodrilus hoffmeisteri* and *Branchyura sowerbyi* (Kikuchi & Kurihara 1977) and in sediment with mixed populations of *L. hoffmeisteri* and *L. cervix* (Fisher 1982). This increased  $\text{NH}_4^+$  might have been due either to animal excretion or to enhanced microbial activity (Matisoff et al. 1985). Certainly, tubificids excrete  $\text{NH}_4^+$  at quite high rates,  $0.6 \text{ nmol NH}_4^+ \text{ ind.}^{-1} \text{ h}^{-1}$  (Gardner et al. 1983), and this occurs well below the sediment surface (Wisniewski & Planter 1985). It is probable that most of the increased  $\text{NH}_4^+$  in our sediments could be attributed to direct worm excretion, rather than to increased microbial activity. Increases in porosity and/or diffusivity at the sediment surface due to *Tubifex tubifex* may lead to enhanced  $\text{NH}_4^+$  fluxes out of the sediment, compared to non-inhabited microcosms. Our measurements showed indeed increased  $\text{NH}_4^+$  fluxes in the presence of the worms.

Nitrification rates increased by 23% from 0 to  $20000 \text{ ind. m}^{-2}$ , and decreased by 30% from  $20000$  to  $70000$



ind. m<sup>-2</sup>. This accords well with increased rates of nitrification as organic degradation increases up to the highest organic loading rates, at which an inhibition of nitrification took place (Blackburn 1990, Blackburn & Blackburn 1993, Caffrey et al. 1993). As the rate of organic degradation increased, the quantity of NH<sub>4</sub><sup>+</sup> increased, but the availability of O<sub>2</sub> decreased resulting in a compression of the zone of nitrification into a thinner band, closer to the sediment surface. Eventually, even though NH<sub>4</sub><sup>+</sup> concentrations might be high, the rate of loss of NH<sub>4</sub><sup>+</sup> by diffusion exceeds the rate at which it could be oxidized. This scenario fits the observed data: NH<sub>4</sub><sup>+</sup> concentrations increased with worm density and this was coupled to higher rates of O<sub>2</sub> uptake. The NH<sub>4</sub><sup>+</sup> was probably mostly produced by the worms and at least some of the increased O<sub>2</sub> uptake was also attributed to worm uptake. It is likely, however, that O<sub>2</sub> penetration into the sediment surface decreased with increased worm population densities, partially due to the microbial activity in the fecal pellet layer at the sediment surface. The relatively low degree of coupling between nitrification and denitrification (~25%) also suggests that nitrification was close to the sediment-water interface and that nitrate could escape by diffusion into the overlying water. Chaterpaul et al. (1980) found that the presence of *Limnodrilus hoffmeisteri* and *Tubifex tubifex* (12 000 to 16 000 ind. m<sup>-2</sup>) increased sediment nitrification rates by 40% during a substantially longer incubation period.

### CONCLUSION

The presence of *Tubifex tubifex* enhanced O<sub>2</sub> and NO<sub>3</sub><sup>-</sup> uptake, associated with changes in surface sediment diffusivity, due to their feeding behavior and/or to irrigation activity of the worms. Nitrification was stimulated at low worm densities (20 000 ind. m<sup>-2</sup>) but was inhibited at higher densities (70 000 ind. m<sup>-2</sup>). It is likely that the O<sub>2</sub> penetration depth decreased with increasing worm density, due to the transport of reduced compounds and organic matter, with the fecal pellets, to the sediment surface. These anoxic conditions led to increased rates of denitrification and were responsible for the decrease in nitrification at higher worm densities. An average of 25% of the NO<sub>3</sub><sup>-</sup> uptake by the sediment was removed from the system by denitrification. There was some evidence for the reduction of NO<sub>3</sub><sup>-</sup> to NH<sub>4</sub><sup>+</sup> at a population density of 70 000 ind. m<sup>-2</sup> (data not shown). This process was not measured at other worm densities. It is most likely that the discrepancy between the rate of NO<sub>3</sub><sup>-</sup> uptake and the rate of denitrification was due to NO<sub>3</sub><sup>-</sup> assimilation by micro-organisms. The detritus in freshwater systems

often has a high C:N ratio, because of its origins in plant structural material. Micro-organisms utilizing this material can be nitrogen-starved (Fenchel & Blackburn 1979). The higher rates of denitrification were counterbalanced by higher rates of NH<sub>4</sub><sup>+</sup> flux from the sediment, due mainly to worm excretion. It is likely, however, that the presence of bioturbating worms, stimulating rates of denitrification, resulted in a net loss of nitrogen that could otherwise have been used by the primary producers.

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