

# Effects of Ammonia and Ammonium on Tolerance and Byssogenesis in *Perna viridis*

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**ABSTRACT:** Individuals of the commercially important green mussel *Perna viridis* (Kuriakose and Nair, 1976) were exposed to different concentrations of ammonia ( $\text{NH}_3$ ) and ammonium ( $\text{NH}_4^+$ ), to determine their reactions with reference to tolerance and byssogenesis. Eight concentrations of ammonia, ranging from 2.75 to 33.00  $\text{mg l}^{-1}$ , and 7 concentrations of ammonium, ranging from 5.00 to 75.00  $\text{mg l}^{-1}$ , were tested. Young individuals in the size range 20 to 24 mm shell length, collected from an unpolluted rocky shore, were used as test organisms. The 96-h lethal concentration causing 50 % mortality (LC 50) was 7.60  $\text{mg l}^{-1}$  for  $\text{NH}_3$  and 13.00  $\text{mg l}^{-1}$  for  $\text{NH}_4^+$ . For ammonia, the difference between LC 50 values at 48 and at 96 h is drastic. In sublethal concentrations *P. viridis* retarded or suspended pedal-gland activity. A sharp reduction in the number of threads produced occurs at 5.50 and 8.25  $\text{mg l}^{-1}$  of ammonia. At higher concentrations the byssus threads secreted displayed reduced rates of tanning and hardening. The difference in the number of byssus threads produced at various ammonia concentration was highly significant. Compared to controls the discs developed at the tips of the byssus threads did not possess normal adhesive properties.

## INTRODUCTION

This paper is part of an investigation of the ecology of waters off Mangalore (Arabian Sea), which receive effluents comprising mainly urea and ammonia from a fertilizer complex. *Perna viridis* (Syn. *Mytilus viridis*) forms extensive beds on the rocky shores along the coast of Mangalore. This species is regularly exploited by local fishermen. The analysis of its tolerance to various concentrations of ammonia and their effect on byssogenesis can help to elucidate potential influences of ammonia on essential life processes of this commercially important bivalve.

Information on the effect of  $\text{NH}_3$  and  $\text{NH}_4^+$  on the life processes of marine organisms is scarce (for a review consult 'Marine Ecology' Volume III: Kinne, 1976, pp. 80–93). Working on the effects of ammonium nitrate on fertilization and early development of *Chromytilus meridionalis* (Brown, 1974) found that concentrations as low as 0.5 ppm retard early development. Currie et al. (1974) obtained similar results. Brown and Currie (1973) determined the tolerance of *Bullia digitalis* to solutions of  $\text{NH}_4$   $\text{NO}_3$  in natural sea water.

## MATERIAL AND METHODS

The green mussel *Perna viridis* (Kuriakose and Nair, 1976) was collected from an unpolluted region at

Someswara rocky shore (12°47'N; 74°51'E). All individuals were obtained during low tide from the subtidal belt. They were transported to the laboratory and kept unfed in large polythene trays, in aerated sea water for 24 h before commencement of the experiments. Young mussels in the size range 20 to 24 mm were used for the experiments. In a single set of experiments only members of the same population were examined.

Liquor ammonia, containing 25 % free ammonia was the source of molecular ammonia ( $\text{NH}_3$ ) and ammonium sulphate was the source of ammonium ( $\text{NH}_4^+$ ). Mortality tests were conducted over 96 h. Inability to close the valves upon mechanical stimulus and valve gaping of 5 mm, were the criteria used to define death. To check the possibility of revival, those mussels considered dead by the above criteria were exposed to normal control conditions for 24 h. Dead individuals were removed from the experimental media at 12 h intervals. The experimental vessels were cylindrical glass troughs of 5 l capacity containing 4 l filtered sea water. Ten mussels were exposed to each concentration (Table 2 and 3). Byssus-thread formation was calculated as the number of threads secreted by one mussel. Readings were taken at 12 h intervals (at 0600 and 1800 h). Since the majority of the threads formed developed adhesive discs, these discs were counted and then the whole byssal mass (stem and threads) was cut-off flush with the valves, and the

mussels were left in the test solution. In assessing the percentage of mussels which had attached, only the presence of byssus threads with adhesive discs was taken into consideration. It was found that cutting off the threads does not affect further byssogenesis.

Table 1. Concentrations of ammonia ( $\text{NH}_3$ ) in various test solutions and their respective pH values. pH of control sea water = pH 8.05

$\text{NH}_3$ ( $\text{mg l}^{-1}$ )	pH
1.375	8.225
2.750	8.325
5.500	8.475
8.250	8.600
11.000	8.700
16.500	8.850
22.000	9.000
33.000	9.200

## RESULTS

The amount of the molecular form of ammonia increases conspicuously in the pH range of 8.0 to 9.5. The water temperature also plays a marked role in dissociation of  $\text{NH}_3$  from its source. Waters of high temperature and high pH will therefore contain a high amount of the molecular form of ammonia, irrespective of the total ammonia concentration (Fig. 1). During the present study, the pH of the sea water varied between 8.0 and 8.3, the temperature between 28.0 ° and 30.0 °C. The addition of liquor ammonia greatly affected the pH of the test solution. Thus, an increase in  $\text{NH}_3$  concentration from 1.375 to 33  $\text{mg l}^{-1}$  altered the pH from 8.225 to 9.200. The increase in pH was directly proportional to the increase in  $\text{NH}_3$  concentration (Table 1).

## MORTALITY

### Lethal Effects of Ammonia

Death prevailed in 5.5 and 8.25  $\text{mg l}^{-1}$   $\text{NH}_3$  after 48 h exposure. 100 % mortality occurred before 96 h at 16.5 and 22.0  $\text{mg l}^{-1}$ . No mussels kept at 33.0  $\text{mg l}^{-1}$  survived beyond 36 h (Table 2; Fig. 2).

ET 50 values worked out from the results obtained on mortality are presented in Figure 2. It is clear from this

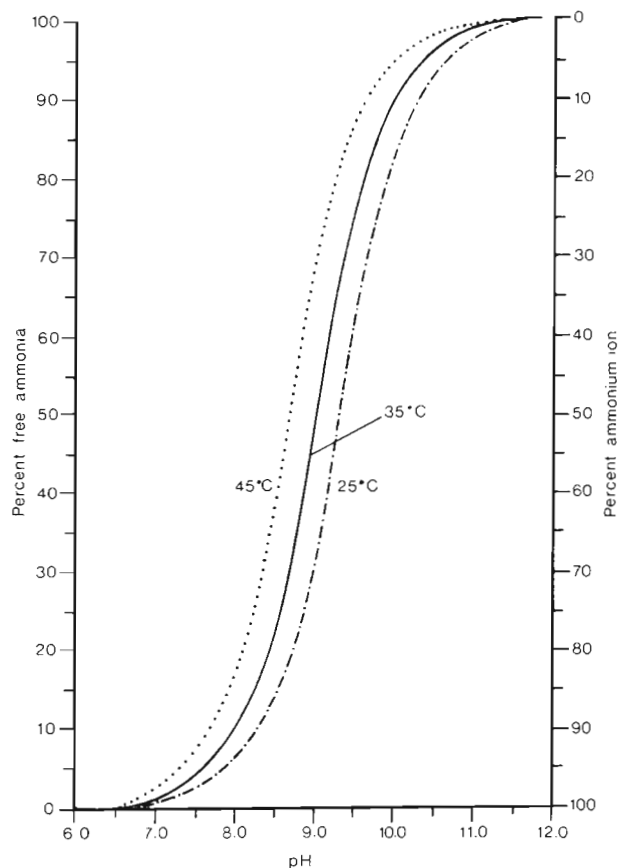


Fig. 1. Per cent free ammonia as a function of pH and temperature. (After Wuhurmann, 1947)

Table 2. *Perna viridis*. Cumulative percentage mortality of individuals exposed to different  $\text{NH}_3$  concentrations

$\text{NH}_3$ Concentration ( $\text{mg l}^{-1}$ )	Time (h)							
	12	24	36	48	60	72	84	96
Control:	0	0	0	0	0	0	0	0
2.75	0	0	0	0	0	0	0	0
5.50	0	0	0	20	20	20	20	20
8.25	0	0	0	20	20	20	40	50
11.00	0	0	0	20	40	40	60	100
13.75	0	0	0	40	40	60	80	100
16.50	0	0	20	40	40	80	100	
22.00	0	40	60	100				
33.00	0	60	100					



figure that a conspicuous difference in time exists for the incidence of 50 % mortality between concentrations of 16.5 and 20.0 mg l<sup>-1</sup>. The lethal concentration causing 50 % mortality (LC 50) in a given length of time values were derived for different concentrations. Figure 2 shows that the 48-h LC 50 is 14.0 mg l<sup>-1</sup> the

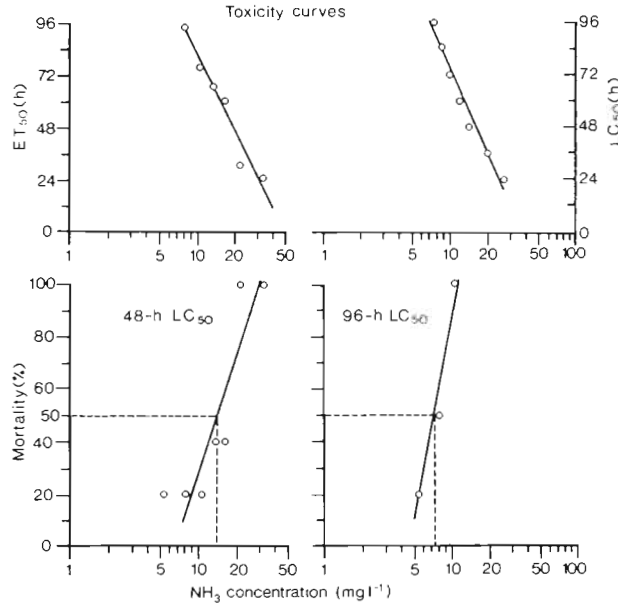


Fig. 3 *Perna viridis*. Median lethal concentrations (LC 50) for different NH<sub>3</sub>-exposure times and toxicity curves

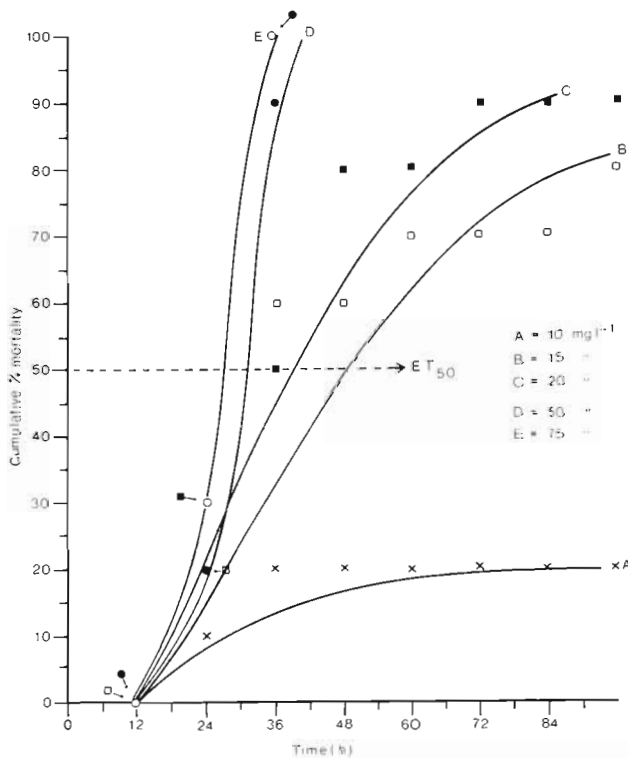


Fig. 4. *Perna viridis*. Mortality of individuals exposed to different NH<sub>4</sub><sup>+</sup> concentrations

96-h LC 50, 7.6 mg l<sup>-1</sup> (Fig. 3). The 5 % confidence limits varied between 13.0 and 15.5 mg l<sup>-1</sup> for 48 h and between 7.2 and 7.8 mg l<sup>-1</sup> for 96 h.

**Lethal Effects of Ammonium**

Ammonium sulphate at various concentrations was used so as to derive NH<sub>4</sub><sup>+</sup> values ranging from 5.0 to 75.0 mg l<sup>-1</sup>. A slight reduction in pH occurred upon addition of (NH<sub>4</sub>)<sub>2</sub> SO<sub>4</sub>; pH of the media ranged from 8.125 (5.0 mg l<sup>-1</sup>) to 7.85 (75.0 mg l<sup>-1</sup>).

Mortality data are presented in Table 3. 80 and 90 % of the test individuals were killed after 96 h exposure in 15.0 and 20.0 mg l<sup>-1</sup> NH<sub>4</sub><sup>+</sup>, respectively. 100 % mortality occurred in 48 h at 50.0 mg l<sup>-1</sup> and in 36 h, at 75.0 mg l<sup>-1</sup>. The ET 50 values, computed from mortality rates, show that 50 % of the test organisms died within 49.8 h at 15.0 mg l<sup>-1</sup>, whereas 50 % died after 30 h at 75.0 mg l<sup>-1</sup>. Curiously, the difference between the ET 50 values for 50.0 and for 75.0 mg l<sup>-1</sup> is only 1.8 h (Fig. 4). The LC 50 values were derived from the mortality rates. The 48-h LC 50 was 15.5 mg l<sup>-1</sup>; the 96-h LC 50, 13.0 mg l<sup>-1</sup>. The 5 % confidence limits for the duration of 48 and 96 h range from 13.75 to 16.00 mg l<sup>-1</sup> and 12.50 and 14.00 mg l<sup>-1</sup>, respectively (Fig. 5).

**BYSSOGENESIS**

The process of byssus-thread secretion by some Bivalvia is referred to as byssogenesis. The rate of production, the nature of secretion, the structure of

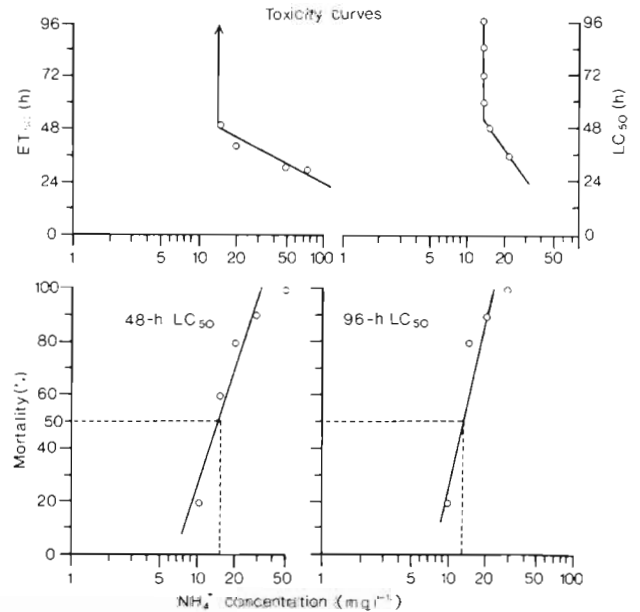


Fig. 5 *Perna viridis*. Median lethal concentrations (LC 50) for different NH<sub>4</sub><sup>+</sup>-exposure times and toxicity curves





lead to hyperplasia of the gill epithelium. According to Currie et al. (1974),  $\text{NH}_4\text{NO}_3$  concentrations up to 10 ppm. have no significant effect on fertilization of *Chromytilus meridionalis*, although embryonic development was retarded, even at values as low as 0.5 ppm. The asymptotic pattern of the ET 50 curve indicates that low  $\text{NH}_4^+$  concentrations do not cause lethal effects, indicating that  $\text{NH}_4^+$  is less toxic than  $\text{NH}_3$ .

Where *Perna viridis* has survived ammonia and ammonium exposure, byssus-thread formation tended to be affected negatively and the pedal activity dropped in concentrations just above the threshold levels. The decrease in the activity is more conspicuous in  $\text{NH}_3$  than  $\text{NH}_4^+$  exposed individuals.

The site of  $\text{NH}_3$  and  $\text{NH}_4^+$  entry could be the mantle, gill, gut or the adductor muscle. In general, the rate of  $\text{NH}_3$  entry through the cell wall is quicker than that of  $\text{NH}_4^+$ . Due to increased dissociation of  $\text{NH}_3$  at pH values above 8.2, maintenance of high pH in the sea water bathing the mantle and gills of *P. viridis* during exposure, will result in an increased  $\text{NH}_3$  entry into the tissue. *P. viridis* is found to thrive well in sea water where the pH ranges from 7.8 to 8.6 (Rao and Menon, in press). Hence it may be assumed that, within the above range, pH is not a limiting factor to normal life processes of *P. viridis*. However, high pH resulting from the presence of  $\text{NH}_3$  might lead to cellular damage.

At higher concentrations the byssus threads of *Perna viridis* showed variations in morphology. The most conspicuous changes were lack of hardening and inability of the adhesive disc to attach firmly. The rate of byssus-thread production is influenced by various environmental factors. These include water movement, salinity, the position occupied by the mussels in the intertidal region, dissolved oxygen concentration, temperature, oil, detergents and mercury salts (Reish and Ayers, 1968; van Winkle, 1970; Roberts, 1975). Improper development of adhesive discs under ammonia stress results in the mussels losing the capacity to attach byssally. During darkness, *Perna viridis* secretes more byssus threads than when exposed to light.

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