

Microbial regeneration of ammonium in the water column of Davies Reef, Australia*

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ABSTRACT: Ammonium regeneration by microheterotrophs was measured during winter in the water column of Davies Reef, on the Great Barrier Reef. Regeneration rates were among the lowest reported, ranging from undetectable to $0.0112 \mu\text{mol NH}_4^+ \text{-N l}^{-1} \text{ h}^{-1}$. Regeneration was highest inside a patch reef (0.0112), followed by reef flat (0.0050), lagoon (0.0017) and fore-reef ($0.0013 \mu\text{mol NH}_4^+ \text{-N l}^{-1} \text{ h}^{-1}$). Bacteria and heterotrophic nanoplankton accounted for 40 to 88 % of the NH_4^+ regenerated in 208 μm filtered water. There was a trend of increasing importance of these organisms as water passed from the fore-reef, across the reef flat, and into the lagoon and patch reef areas. Remineralization was highest where water was in most intimate contact with the reef benthos, which may reflect a response of the pelagic community to the input of organic matter (e.g. mucus) to the water column by the reef benthos. A first approximation of the relative importance of bacteria versus heterotrophic nanoplankton in NH_4^+ regeneration, established from measures of biomass and growth rates for both groups, suggested that suspended bacteria were about an order of magnitude more important than nanoplankton. Pelagic ammonium regeneration supplied almost 3 times more inorganic N to the water column in lagoonal portions of the reef than sediment nutrient regeneration did. Only over the reef flat was pelagic NH_4^+ regeneration sufficient to fully meet calculated phytoplankton N demands. Long biologically-mediated turnover times of NH_4^+ (21 to 154 h) suggest a greater importance of physical rather than biological processes in the water column of this reef during winter.

INTRODUCTION

Nutrient requirements of neritic primary producers are met through a combination of inputs from fresh-water runoff, coastal upwelling, deep water intrusions and local pelagic or benthic nutrient regeneration. Coral reefs away from continental margins generally exist in oligotrophic, nutrient-poor regions and the input of inorganic nutrients from the surrounding milieu is frequently accepted as being relatively unimportant (Johannes et al. 1972). With respect to ammonium – the preferred and frequently most limiting nutrient species to the gross primary production of phytoplankton (Ryther & Dunstan 1971, McCarthy et al. 1977) – the relative importance of sediment and pelagic regeneration has not been determined. In coral

reefs (Kinsey 1985a, Ducklow 1987) and shallow water systems in general (Hargrave 1973), most metabolism is mediated by the benthos and it might be expected that ammonium regeneration in the sediments largely dictates nutrient concentrations in the overlying water column. However, as water approaches, passes over reef crest and reef flat regions, and exits from the back reef, microbial activity in the water column appears to be greatly stimulated in response to utilizable substrates released from the reef community (Johannes 1967, Ducklow 1987). The living and detrital organic nitrogen released by the reef community is remineralized to ammonium and CO_2 via microzooplankton catabolism and excretion, and by the metabolic processes of the pelagic, heterotrophic microbial community (Dugdale & Goering 1967, Harrison 1980).

In this paper, we present the first direct estimates of ammonium regeneration by microheterotrophs in the water column at Davies Reef, on the Great Barrier Reef. Also reported are estimates of the relative importance

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of benthic versus pelagic ammonium regeneration, production rates of pelagic bacteria and heterotrophic nanoplankton, the relative importance of bacteria versus heterotrophic nanoplankton and microplankton in regenerating inorganic nitrogen, and the spatial variability in pelagic ammonium regeneration in a coral reef system.

DESCRIPTION OF THE AREA

Davies Reef, a platform reef of about 5.8 by 2.6 km ($18^{\circ}51'S$; $147^{\circ}39'E$), is on the inner edge of the main reef tract on the eastern half of the continental shelf off Townsville, Australia (Fig. 1). The reef is partially surrounded by other coral reefs separated by water in excess of 60 m depth. The windward face of the reef comprises a solid wall with extensive cover of corals plus other invertebrates and algae, especially *Halimeda* spp. Behind the fore-reef slope is a reef flat of mixed substratum and increasing live coral cover to leeward. A distinct algal ridge or rubble bank is absent from this reef. The leeward side is a broken wall with channels and actively growing areas. The lagoonal system is generally 10 to 25 m deep with isolated patch reefs (bommies) which increase in size at the back of the reef (northwest face). During August 1984, water temperature averaged 22 to 24°C, salinity 35.3 ppt, tidal range 1 to 3.5 m (neap/spring), and day length 11.5 h. Winds were generally light and variable but there were periods when they exceeded 10 m s^{-1} . During sampling, winds ranged from near calm to 3 m s^{-1} . Water residence time ranged from less than 30 min on the reef flat, up to about 12 to 48 h in the East lagoon (Kinsey 1985b, Pickard 1986, MECOR unpubl. data – Australian Institute of Marine Sciences, Townsville, Queensland).

MATERIALS AND METHODS

Pelagic ammonium remineralization was investigated during August 1984 at 4 locations along a transect extending from the seaward side of the fore-reef to the rear of Davies Reef and running mostly parallel to the direction of net water flow (Fig. 1). As one of our sampling objectives was to determine whether significant differences in growth rate and remineralization by the heterotrophic microbial community in the water column could be detected as water flowed over various benthic subsystems of the coral reef, we chose sites extending across the entire reef system. Site 1 was a fore-reef site in 57 m of water approximately 100 m in front of the reef crest. Site 2 was at the downstream edge of an approximately 300 m wide reef flat region

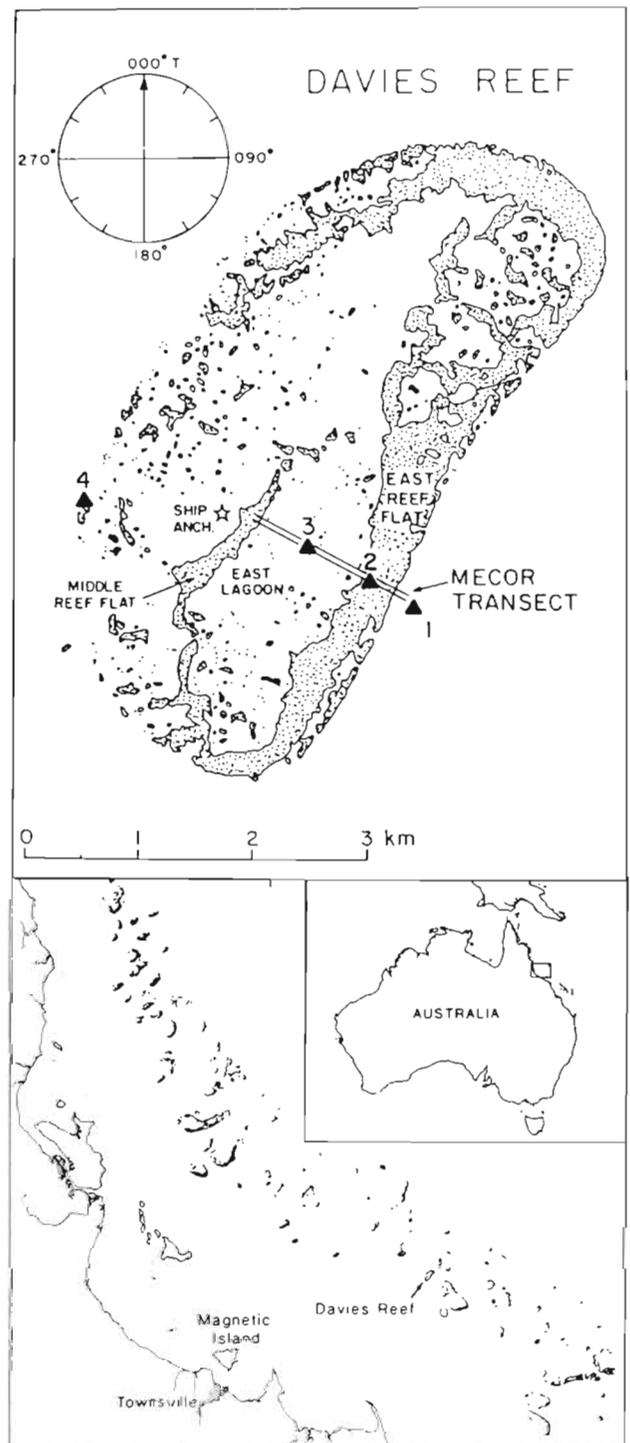


Fig. 1. Location of sampling sites at Davies Reef, central Great Barrier Reef, off Townsville, Australia

(exposed only during very low tides). Site 3 was in 15 m of water in the center of the East (front) lagoon at Davies. Site 4 was internal to a large (approximately 60 by 30 m) cavernous patch reef at the rear of the reef. Rhodamine dye injections indicated that the cave sam-

pled (approximately 1 m internal diameter by 15 m long) had sluggish unidirectional flow down from the top of the patch reef flat and out of the rearward face of the patch reef. At Sites 1, 2 and 3, water was collected with 5 or 10 l Niskin bottles at mid water column depth. From near center of the patch reef cave, water was collected with a peristaltic pump through 2 mm ID Tygon brand vinyl tubing (Fisher Scientific, Springfield, New Jersey, USA) attached to a bamboo spar in a manner similar to that of Andrews & Muller (1983). All sample incubations were performed aboard a research vessel anchored in the lagoon.

For each water mass sampled, ammonium regeneration assays were run using 208 μm and 15 μm screened water. Size fractionation was accomplished by reverse-filtration. Within 30 min of collection, water for ammonium regeneration experiments was gently transferred through 208 μm screening to a 12 l polypropylene bucket. While bucket contents were slowly stirred, 2.4 l portions were siphoned into two 2.5 l glass reagent bottles. A PVC tube (20 cm ID) with 15 μm nitex screening (DuPont woven nylon monofilament – Tetko Inc., Elmsford, New York, USA) attached across the bottom was allowed to sink under its own weight (<2.5 cm head) into the water remaining in the bucket. Reverse-filtered water was gently siphoned from inside the tube into two 2.5 l glass bottles.

Sufficient $\text{N}^{15}\text{-(NH}_4\text{)}_2\text{SO}_4$ (99 %) was added to each bottle to make a final concentration of 0.1 μM ^{15}N . Immediately after tracer addition, each bottle was swirled and an 800 ml sample withdrawn. The bottles were then placed in an on-deck flowing seawater incubator covered with neutral density screening allowing 50 % light transmission. From the 800 ml samples, 300 ml were used to rinse filtering apparatus (including Gelman glass fiber GF/F) and polyethylene storage bottles and then discarded. Another 100 ml were used for zero-time ammonium analysis and the remainder of the filtrate was frozen until further processing on shore. Bottles were resampled 1 and 3 h after tracer addition.

Ammonium concentration was redetermined following thawing and ammonium stripped from stored samples within 5 d of collection at the onshore laboratory. Prior to stripping, 1.5 μmol $^{14}\text{N}\text{-(NH}_4\text{)}_2\text{SO}_4$ carrier were added to the 400 ml sample. The solvent extraction procedure described by Dudek et al. (1986) was used to strip ammonium nitrogen for the determination of relative ^{15}N abundance. In this procedure ammonium is converted to indophenol using a modification of the phenol-hypochlorite reaction for seawater ammonium analysis. Sodium aquopentacyanoferrate is used rather than sodium nitroprusside for catalyzing the reaction. The indophenol is extracted into methylene chloride, concentrated by partial evaporation of

the solvent and dried on a glass fiber filter (Whatman 934 AH). Particular care was taken to ensure that sample pH was lowered to the optimum level ($\sim\text{pH}$ 6.0), and that vacuum pressure (700 mm Hg) and water bath temperature (35 °C) were precisely controlled during concentration. Filters were dried at 80 °C and stored in plastic scintillation vials. Upon return to the USA, vials with filters were stored under vacuum until ^{15}N analysis.

^{15}N content was analyzed by emission spectrometry following a modification of the micro-Dumas procedure (Dudek et al. 1986). Filters were ground with 0.5 g precombusted (500 °C) cupric oxide wire (Fisher Scientific, Springfield, New Jersey, USA) and stored in evacuated vacutainers (vacuum-tight test tubes – Becton-Dickinson, Rutherford, New Jersey, USA) until placement into a 5 mm OD pyrex glass discharge tube containing about 2 cm precombusted (900 °C) CaO. The tubes were evacuated to $<10^{-5}$ Torr ($=1.3 \times 10^{-3}$ Pa), sealed, combusted for 8 h at 500 °C and analyzed on a Jasco emission spectrometer.

A modified Strickland & Parsons (1972) procedure was employed for determining ammonium concentration. Sodium nitroprusside was recrystallized prior to reagent preparation in order to lower blank values. Absorbances were read at 630 nm in 10 cm path-length cells. All glassware used in these analyses was well aged, combusted at 450 °C, soaked in 10 % HCl for at least 4 h and rinsed thoroughly with deionized water before use. Incubation bottles were rinsed further with the appropriately size-fractionated seawater. Sample bottles and separatory funnels were further cleansed of ammonium by adding reagents immediately prior to use (burnt out).

Ammonium regeneration rates were calculated from measurements of ammonium concentration and isotope ratio according to the Blackburn (1979) equations (Laws 1984). Corrections to the measured isotope ratio were made for isotope contamination during sample freezing, ammonium extraction, and micro-Dumas combustion.

Estimates of bacterial and microflagellate biomass and production were made as described in Ducklow & Hill (1985a, b) and Sherr et al. (1984, 1986).

RESULTS AND DISCUSSION

During the sampling period, ammonium standing stock concentrations ranged from 0.07 to 0.37 μM throughout the water column at Davies Reef (Table 1). Although sample variability over depth at any one site was very low (average coefficient of variation 0.03 μM), temporal and horizontal spatial variability were significant (coefficient of variation 0.47 μM). There were no

Table 1. Concentrations of ammonium in the water column at several sites within Davies Reef. All samples were collected in August 1984. Units are μM ; values in parentheses are standard deviations

Location	Mean concentration	No. of samples	Range
Fore-reef	0.201 (.05)	6	0.12–0.25
Reef flat	0.174 (.06)	6	0.10–0.27
Lagoon	0.185 (.11)	14	0.07–0.37
Patch reef			
Internal	0.24	2	0.21–0.27
External	0.14	2	0.13–0.14

clear cut diel or horizontal trends apparent in the distribution of ammonium with the exception of the patch reef, where concentrations were higher inside the patch reef structure than they were in the surrounding water. Andrews & Muller (1983) noted a similar pattern for nitrate in patch reefs at Davies Reef. The absence of spatial patterns within the overall system indicates no discernible effect on ammonium concentrations due to passage of water over the reef benthos. This is in contrast with the frequent observation of net export of NO_3^- from coral reefs (Webb et al. 1975, Andrews & Muller 1983). The distribution of ammonium is also in contrast with the pattern Kinsey observed (1985b, unpubl. MECOR data) of higher NO_3^- concentrations in water exiting the East lagoon region of Davies Reef.

Remineralization rates of ammonium in the water column over Davies Reef ranged from undetectable to $0.0112 \mu\text{mol NH}_4^+ \cdot \text{N l}^{-1} \text{ h}^{-1}$ (Table 2). There were considerable temporal and spatial differences in rates. Temporal variation in estimates of remineralization in the lagoon (ranging from undetectable to $0.0025 \mu\text{mol NH}_4^+ \cdot \text{N l}^{-1} \text{ h}^{-1}$) appeared to be random and not related to tidal or diel rhythms. Variation of this magnitude may reflect a heterogeneous distribution of microbial and metazoan biomass in the water column at Davies Reef. Ammonium remineralization rates, averaged by reef region, spanned 2 orders of magnitude: highest in the patch reef ($0.0112 \mu\text{mol NH}_4^+ \cdot \text{N l}^{-1} \text{ h}^{-1}$) followed by

Table 2. Mean rates of pelagic ammonium regeneration at Davies Reef during August 1984. Units are $\mu\text{mol NH}_4^+ \cdot \text{N l}^{-1} \text{ h}^{-1}$

Site	Remineralization		% of total due to organisms < 15 μm
	< 208 μm	< 15 μm	
Fore-reef	0.0013	0.0005	40 %
Reef flat	0.0050	0.0033	66 %
Lagoon	0.0017	0.0015	88 %
Patch reef (internal)	0.0112	0.0080	71 %

the reef flat ($0.0050 \mu\text{mol NH}_4^+ \cdot \text{N l}^{-1} \text{ h}^{-1}$), the lagoon ($0.0017 \mu\text{mol NH}_4^+ \cdot \text{N l}^{-1} \text{ h}^{-1}$) and the fore-reef ($0.0013 \mu\text{mol NH}_4^+ \cdot \text{N l}^{-1} \text{ h}^{-1}$). Remineralization was highest in those areas where water was in most intimate contact with the reef benthos (shallow-water regions with a high proportion of benthic surface area to water volume). This may reflect a response of the pelagic community to the input of organic matter (e.g. mucus) to the water column by the reef benthos (Paul et al. 1986, Ducklow 1987).

Bacteria and heterotrophic nanoplankton, operationally defined as organisms < 15 μm , accounted for 40 to 88 % of the NH_4^+ remineralized in 208 μm filtered water at Davies Reef (Table 2). There was a trend of increasing importance of these organisms as water passed from the fore-reef, across the reef flat, into the lagoon and patch reef areas. Again, this may be a result of metabolic stimulation of the microbial community (biomass of which was spatially constant) due to the release of soluble organics such as mucus from the coral community (Paul et al. 1986, Ducklow 1987). It could also be due to removal of 15 to 208 μm organisms by coral predation. There has been increasing evidence for the importance of plankton smaller than 208 μm in remineralizing NH_4^+ in marine systems. Johannes (1964) was the first to demonstrate that microplanktonic excretion rates could be substantial even though their biomass was low. Likewise, Harrison (1978), Caperon et al. (1979), and Paasche & Kristiansen (1982) demonstrated the importance of the small size classes of heterotrophic plankton in ammonium remineralization in a number of estuarine and shelf environments. Glibert (1982) and Harrison et al. (1983) documented the importance of the smallest size classes. In both the Chesapeake Bay and the Sargasso Sea, Glibert (1982) showed that organisms less than 10 μm often contributed up to 100 % of the NH_4^+ remineralized.

A first approximation of the relative importance of bacteria and heterotrophic nanoplankton in pelagic ammonium remineralization can be established from biomass and growth data of both groups (Table 3). Bacterial ammonium remineralization was estimated to be approximately 0.0007 or $0.0044 \mu\text{mol N l}^{-1} \text{ h}^{-1}$, depending on the choice of methods for measuring bacterial production (eukaryotic inhibitors and cell counts or ^3H -thymidine incorporation rate into DNA). In our experiments, the former methods gave higher production estimates, even when we used a thymidine conversion factor of 4×10^{18} cells mole $^{-1}$ (Ducklow & Hill 1985b).

Remineralization by heterotrophic nanoplankton was much smaller and, as for bacteria, varied depending on the method of calculation. Estimates calculated from biomass data were slightly higher than those determined from growth/respiration rates. The mean mea-

Table 3. Estimated ammonium remineralization by bacteria and heterotrophic nanoplankton (HNAN). Bacterial biomass from Ducklow (unpubl.) and Sherr (unpubl.) direct cell counts (Hobbie et al. 1977), assuming 11 fg C per bacterial cell (Bratbak & Dundas 1984). Bacterial production (growth) from unpubl. data sets of Ducklow (^3H -thymidine incorporation into DNA: Ducklow & Hill 1985b) and Sherr (time course cell counts with eukaryotic inhibitors: Sherr et al. 1986). HNAN production from Sherr (unpubl. protozoan grazing data: see Sherr et al. 1984). Respiration calculated from growth assuming 33% growth efficiency. Remineralization estimated from (1) respiration and Redfield stoichiometry and (2) from Sherr et al. (1983) excretion rates per unit flagellate biomass. Measured remineralization is that measured in 10 μm screened water

	Bacteria	HNAN
Measured	2.0 $\mu\text{g C l}^{-1}$	0.5 $\mu\text{g C l}^{-1}$
Biomass	0.0264–0.175 $\mu\text{g C l}^{-1} \text{ h}^{-1}$	0.0042 $\mu\text{g C l}^{-1} \text{ h}^{-1}$
Growth	0.0528–0.35 $\mu\text{g C l}^{-1} \text{ h}^{-1}$	0.0083 $\mu\text{g C l}^{-1} \text{ h}^{-1}$
Respiration (est.)		
Estimated NH_4^+ remineralization		
From respiration	0.0007–0.0044 $\mu\text{mol l}^{-1} \text{ h}^{-1}$	0.0001 $\mu\text{mol l}^{-1} \text{ h}^{-1}$
From biomass	Conversion unknown	0.00025 $\mu\text{mol l}^{-1} \text{ h}^{-1}$
Measured remineralization	0.0018 $\mu\text{mol N l}^{-1} \text{ h}^{-1}$ (bacteria and HNAN)	

sured rate of remineralization at Davies Reef (excluding reef interstices) by plankton passing a 15 μm screen was 0.0018 $\mu\text{mol N l}^{-1} \text{ h}^{-1}$ and is within the range calculated for both bacteria and HNAN. These results suggest that the suspended bacteria at Davies Reef were more important as nutrient remineralizers than the heterotrophic nanoplankton at the time of our sampling. The C/N ratio of the majority of the organic matter microbially degraded in the water column of this coral reef must have been similar to the Redfield ratio and roughly equivalent to the C/N ratio of bacterial protoplasm, otherwise there would not have been such close agreement between measured and calculated rates of ammonium remineralization. In such a situation bacteria must release nitrogen to the environ-

ment in order to maintain the proper C/N ratios within their cells (Caron & Goldman in press). Hopkinson et al. (unpubl.) suggested that nanoplankton assume a greater importance than bacteria in N remineralization only when the C/N ratio of the organic substrates being decomposed exceeds 11:1.

Ammonium regeneration rates observed at Davies Reef are among the lowest reported in the literature (Table 4). They more closely resemble rates observed in the open ocean than in relatively shallow coastal and estuarine areas. This may be a consequence of the very low levels of microbial biomass and production we observed at Davies Reef (Table 2, see also Ducklow 1987). Pelagic ammonium regeneration rates measured in 2 other 'live' bottom, benthic-dominated systems,

Table 4. Rates of pelagic ammonium regeneration in several marine environments ($\mu\text{mol NH}_4^+-\text{N l}^{-1} \text{ h}^{-1}$)

Environment	Site	Rate (range or mean)	Source
Estuarine	Laguna de Terminos, Mexico	0.064–0.409	Hopkinson et al. 1987
	Chesapeake Bay, USA	0.11–1.195	Glibert et al. 1982
	Oslofjord, Norway	0.012–0.061	Paasche & Kristiansen 1982
	Kaneohe Bay, Hawaii	0.010–0.162	Caperon et al. 1970
Coastal	Vineyard Sound, Massachusetts, USA	~0–0.31	Glibert 1982
	Georgia Bight, USA	0.11	Hanson & Robertson unpubl.
	Bedford Basin, Nova Scotia, Canada	0.056	La Roche 1983
	CEPEX, British Columbia, Canada	0.023–0.051	Harrison 1978
	Grays Reef, Georgia Bight, USA	0.048–0.089	Fallon & Hopkinson 1986
	Southern California Bight, USA	0.0006–0.0091	Harrison 1978
	Middle Atlantic Bight	0.008–0.029	Harrison et al. 1983
Davies Reef	0.0013–0.0112	This study	
Open ocean	Scotia Sea	~0–0.13	Glibert 1982
	Sargasso Sea	0.0020–0.0646	Glibert et al. 1982

Gray's Reef in the Georgia Bight and Kaneohe Bay in Hawaii, are approximately 20 times higher than those measured at Davies Reef. The extremely low rates of pelagic remineralization at Davies Reef reflect the low levels of pelagic primary production and the generally oligotrophic nature of the water column adjacent to and over this pristine coral reef system. Rates may be higher during summer when the water resides longer in the reef system (Kinsey 1985b, Moriarty et al. 1985, Ducklow 1987).

In shallow water systems, benthic metabolism and nutrient regeneration are generally greater than pelagic processes (Hargrave 1973). Live bottom portions of coral reefs frequently experience rates of total community production and respiration in excess of $6 \text{ gC m}^{-2} \text{ d}^{-1}$ (Kinsey 1985a); phytoplankton production is usually trivial and certainly less than 10 % of total production (Lewis 1977). During the MECOR expedition benthic nutrient regeneration was measured only in lagoonal sediments. Although benthic respiration was as high as $1 \text{ gC m}^{-2} \text{ d}^{-1}$ (Hansen, unpubl. MECOR data), the ratio of primary production to respiration was nearly balanced, indicating an active, nutrient-demanding algal community. Consequently there was little net loss or flux of nutrients regenerated in the sediments to the overlying water column (about $0.62 \mu\text{mol N m}^{-3} \text{ h}^{-1}$; Hansen pers. comm.). With respect to supplying inorganic nitrogen to the water column in lagoonal portions of Davies Reef, pelagic N regeneration was almost 3 times more important than sediment nutrient regeneration (averaging $1.7 \mu\text{mol N m}^{-3} \text{ h}^{-1}$).

The quantitative importance of pelagic ammonium regeneration can be partially assessed by calculating the proportion of phytoplankton N demand that is potentially supplied by pelagic regeneration. The estimated phytoplankton requirement of N, calculated from the Redfield ratio and ^{14}C estimates of primary production (Furnas pers. comm.) ranged from 0.0016 to $0.0055 \mu\text{mol NH}_4^+ \text{-N l}^{-1} \text{ h}^{-1}$ (Table 5). In deep water

Table 5. Importance of pelagic ammonium remineralization to phytoplankton at Davies Reef (units: $\mu\text{mol NH}_4^+ \text{-N l}^{-1} \text{ h}^{-1}$)

Site	N require- ment*	N regenerated	% of requirement
Fore-reef	0.0033	0.0013	39
Reef flat	0.0016	0.0050	313
Lagoon	0.0055	0.0017	31

* Requirement calculated from areal daily carbon production estimates for fore-reef and lagoon sites and volumetric rates for the reef flat, a C:N uptake ratio of 106:16, a 24 h interval for N uptake, and depths of 60, and 20 m for fore-reef and lagoon sites, respectively

sites outside and in Davies Reef, pelagic ammonium regeneration supplied only about a third of phytoplankton requirements. Given the uncertainties of sampling, analysis, Redfield ratio deviations and conversion factors, it is hard to determine whether this deficit is significant or real. It suggests that perhaps macrozooplankton are also important agents of N recycling in deep waters of Davies Reef. In the shallow reef flat area, however, considerably more N was regenerated by organisms $< 208 \mu\text{m}$ than was needed by the phytoplankton drifting over the flat. Although the water residence time over the reef flat precluded the identification of an upstream-downstream NH_4^+ concentration gradient on the reef flat, it appears as if some inorganic nitrogen was being made available for export to the adjacent lagoonal system.

On the basis of pelagic ammonium remineralization and standing stock concentrations, turnover times of ammonium in the water column at Davies Reef ranged from 21 to 154 h. Long ammonium turnover times relative to estimated water residence times (Kinsey pers. comm., Pickard 1986, and pers. comm.) suggest a greater importance of physical rather than biological processes in the water column of Davies Reef during the winter when we sampled.

A number of factors, including low rates of pelagic ammonium remineralization, low phytoplankton productivity, short water residence times, a lack of nutrient concentration gradients, and high rates of benthic and total system metabolism, demonstrate the almost trivial importance of heterotrophic microbial processes occurring in the water column of Davies Reef during the period of observation (Ducklow 1987). The general description of a coral reef system (Odum 1971) is of high gross productivity, low net ecosystem production, highly conservative nutrient behavior and benthic community dominance. Our work on pelagic ammonium recycling fully supports this description. We found no evidence of net nutrient uptake or release from Davies Reef. We did find evidence, however, that pelagic processes were stimulated by contact with the benthos and that there might be intrasystem transfers of ammonium within Davies Reef.

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